

Ionic and Covalent Naming Practice **Answers**

For each of the following questions, determine whether the compound is ionic or covalent and name it appropriately.

- 1) Na_2CO_3 **sodium carbonate**
- 2) P_4O_{10} **tetraphosphorus decaoxide**
- 3) N_2O_5 **dinitrogen pentoxide**
- 4) FeSO_4 **iron (II) sulfate**
- 5) SiO_2 **silicon dioxide**
- 6) FeCl_3 **iron (III) chloride**
- 7) CoBr_2 **cobalt (II) bromide**
- 8) B_2H_4 **diboron tetrahydride**
- 9) CO **carbon monoxide**
- 10) PCl_3 **phosphorus trichloride**

For each of the following questions, determine whether the compound is ionic or covalent and write the appropriate formula for it.

- 11) dinitrogen trioxide **N_2O_3**
- 12) nitrogen dioxide **NO_2**
- 13) sulfur pentoxide **SO_5**
- 14) lithium acetate **$\text{LiC}_2\text{H}_3\text{O}_2$**
- 15) phosphorus trifluoride **PF_3**
- 16) vanadium (V) oxide **V_2O_5**
- 17) aluminum hydroxide **$\text{Al}(\text{OH})_3$**
- 18) zinc sulfide **ZnS**
- 19) silicon tetrafluoride **SiF_4**
- 20) silver phosphate **Ag_3PO_4**

Ionic Covalent Bonding Test Study Guide Answers

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Table of Contents Ionic Covalent Bonding Test Study Guide Answers

1. Understanding the eBook Ionic Covalent Bonding Test Study Guide Answers
 - The Rise of Digital Reading Ionic Covalent Bonding Test Study Guide Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Ionic Covalent Bonding Test Study Guide Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Ionic Covalent Bonding Test Study Guide Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Ionic Covalent Bonding Test Study Guide Answers
 - Personalized Recommendations
 - Ionic Covalent Bonding Test Study Guide Answers User Reviews and Ratings

- Ionic Covalent Bonding Test Study Guide Answers and Bestseller Lists
- 5. Accessing Ionic Covalent Bonding Test Study Guide Answers Free and Paid eBooks
 - Ionic Covalent Bonding Test Study Guide Answers Public Domain eBooks
 - Ionic Covalent Bonding Test Study Guide Answers eBook Subscription Services
 - Ionic Covalent Bonding Test Study Guide Answers Budget-Friendly Options
- 6. Navigating Ionic Covalent Bonding Test Study Guide Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Ionic Covalent Bonding Test Study Guide Answers Compatibility with Devices
 - Ionic Covalent Bonding Test Study Guide Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Ionic Covalent Bonding Test Study Guide Answers
 - Highlighting and Note-Taking Ionic Covalent Bonding Test Study Guide Answers
 - Interactive Elements Ionic Covalent Bonding Test Study Guide Answers
- 8. Staying Engaged with Ionic Covalent Bonding Test Study Guide Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Ionic Covalent Bonding Test Study Guide Answers
- 9. Balancing eBooks and Physical Books Ionic Covalent Bonding Test Study Guide Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Ionic Covalent Bonding Test Study Guide Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Ionic Covalent Bonding Test Study Guide Answers
 - Setting Reading Goals Ionic Covalent Bonding Test Study Guide Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Ionic Covalent Bonding Test Study Guide Answers
 - Fact-Checking eBook Content of Ionic Covalent Bonding Test Study Guide Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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