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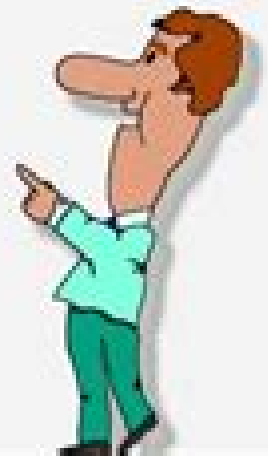
$$C = \frac{n}{V}$$

Molar concentration is the most commonly used in chemistry.

amount of solute (in moles)

Molar concentration = $\frac{\text{-----}}{\text{volume of solution (in litres)}}$

UNITS: (mol/L) or (mol .L⁻¹) or M



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