

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2 \text{MgO}$ **Synthesis**
- 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
- 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al(OH)}_3 + 3\text{KCl}$ **Double Replacement**
- 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
- *5) $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
- 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3 \text{O}_2$ **Decomposition**
- 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
- 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
- 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
- *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2 \text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2 \text{NaBr} \rightarrow 2 \text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg(ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3 \text{O}_2$
- 14) Synthesis: $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
- 15) Double replacement: $3 \text{Ca(OH)}_2 + 2 \text{FeCl}_3 \rightarrow 2 \text{Fe(OH)}_3 + 3 \text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu(NO}_3)_2 \rightarrow \text{Fe(NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2 \text{Hg}_2\text{O} \rightarrow 4 \text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2 \text{AgNO}_3 \rightarrow 2 \text{Ag} + \text{Cu(NO}_3)_2$ [Copper (II) is used here]

Types Of Chemical Reactions Answers

Eve Hartman, Wendy Meshbesh



Types Of Chemical Reactions Answers

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more substances the reactants are converted to one or more different substances the products Substances are either chemical elements or compounds A chemical reaction rearranges the constituent atoms of the reactants to create different substances as products Common Chemical Reactions is a comprehensive book that describes basic concepts about the conservation of matter and patterns to write chemical reactions The book encloses details about different types of reactions such as combination reactions decomposition reactions combustion reactions acid base neutralization reactions redox reactions and replacement reactions This book describes basic concepts of reaction kinetics parameters for measuring the progress of chemical reactions and variables that affect reaction rates This book is a perfect combination of knowledge and examples about different types of reactions *Chemical Reactions and Chemical Reactors* George W. Roberts, 2008-03-14 Focused on the undergraduate audience Chemical Reaction Engineering provides students with complete coverage of the fundamentals including in depth coverage of chemical kinetics By introducing heterogeneous catalysis early in the book the text gives students the knowledge they need to solve real chemistry and industrial problems An emphasis on problem solving and numerical techniques ensures students learn and practice the skills they will need later on whether for industry or graduate work [A Framework for K-12 Science Education](#) National Research Council, Division of Behavioral and Social Sciences and Education, Board on Science Education, Committee on a Conceptual Framework for New K-12 Science Education Standards, 2012-02-28 Science engineering and technology permeate nearly every facet of modern life and hold the key to solving many of humanity's most pressing current and future challenges The United States position in the global economy is declining in part because U S workers lack fundamental knowledge in these fields To address the critical issues of U S competitiveness and to better prepare the workforce A Framework for K 12 Science Education proposes a new approach to K 12 science education that will capture students interest and provide them with the necessary foundational knowledge in the field A Framework for K 12 Science Education outlines a broad set of expectations for students in science and engineering in grades K 12 These expectations will inform the development of new standards for K 12 science education and subsequently revisions to curriculum instruction assessment and professional development for educators This book identifies three dimensions that convey the core ideas and practices around which science and engineering education in these grades should be built These three dimensions are crosscutting concepts that unify the study of science through their common application across science and engineering scientific and engineering practices and disciplinary core ideas in the physical sciences life sciences and earth and space sciences and for engineering technology and the applications of science The overarching goal is for all high school graduates to have sufficient knowledge of science and engineering to engage in public discussions on science related issues be careful consumers of scientific and technical information and enter the careers of their choice A Framework for K 12 Science Education is the first step in a process that can inform state level decisions and achieve a research grounded basis for improving science instruction and learning across the country The book

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

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