

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ **Synthesis**
- 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
- 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al(OH)}_3 + 3\text{KCl}$ **Double Replacement**
- 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
- *5) $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
- 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ **Decomposition**
- 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
- 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
- 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
- *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2\text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg(ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3\text{O}_2$
- 14) Synthesis: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- 15) Double replacement: $3\text{Ca(OH)}_2 + 2\text{FeCl}_3 \rightarrow 2\text{Fe(OH)}_3 + 3\text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu(NO}_3)_2 \rightarrow \text{Fe(NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2\text{Hg}_2\text{O} \rightarrow 4\text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu(NO}_3)_2$ [Copper (II) is used here]

Classifying Chemical Reactions Worksheet Answer

Ron Darby, Raj P. Chhabra



Classifying Chemical Reactions Worksheet Answer

Chemistry 2e Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition [POGIL Activities for High School Chemistry](#) High School POGIL Initiative, 2012

Balancing Chemical Equations Worksheet Crispin Collins, 2020-09-12 Struggling with balancing chemical reaction Balancing chemical equations can look intimidating for lot of us The good news is that practice makes perfect Master balancing skill with this workbook packed with hundreds of practice problems This book is for anyone who wants to master the art of balancing chemical reactions First few chapters of this book are step by step explanation of the concepts and other chapters are for practicing problems This book help students develop fluency in balancing chemical equation which provides plenty of practice Methods to solve with the explanation Total of 550 problems to solve with answer key 450 chemical reactions to practice with answer key 100 practice problems that are needed before balancing a chemical reaction with answer key Click the Buy now button to take advantage of this book to help yourself in mastering balancing skill *World of Chemistry* Steven S. Zumdahl, Susan L. Zumdahl, Donald J. DeCoste, 2006-08 Our high school chemistry program has been redesigned and updated to give your students the right balance of concepts and applications in a program that provides more active learning more real world connections and more engaging content A revised and enhanced text designed especially for high school helps students actively develop and apply their understanding of chemical concepts Hands on labs and activities emphasize cutting edge applications and help students connect concepts to the real world A new captivating design clear writing style and innovative technology resources support your students in getting the most out of their textbook Publisher [Oxidizing and Reducing Agents](#) Steven D. Burke, Rick L. Danheiser, 1999-07-09 Oxidizing and Reducing Agents S D Burke University of Wisconsin at Madison USA R L Danheiser Massachusetts Institute of Technology Cambridge USA Recognising the critical need for bringing a handy reference work that deals with the most popular reagents in synthesis to the laboratory of practising organic chemists the Editors of the acclaimed Encyclopedia of Reagents for Organic Synthesis EROS have selected the most important and useful reagents employed in contemporary organic synthesis Handbook of Reagents for Organic Synthesis Oxidizing and Reducing Agents

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Chemical Kinetics and Reaction Dynamics Santosh K. Upadhyay, 2007-04-29 Chemical Kinetics and Reaction Dynamics brings together the major facts and theories relating to the rates with which chemical reactions occur from both the macroscopic and microscopic point of view This book helps the reader achieve a thorough understanding of the principles of chemical kinetics and includes Detailed stereochemical discussions of reaction steps Classical theory based calculations of state to state rate constants A collection of matters on kinetics of various special reactions such as micellar catalysis phase transfer catalysis inhibition processes oscillatory reactions solid state reactions and polymerization reactions at a single source The growth of the chemical industry greatly depends on the application of chemical kinetics catalysts and catalytic processes This volume is therefore an invaluable resource for all academics industrial researchers and students interested in kinetics molecular reaction dynamics and the mechanisms of chemical reactions

Study Guide/Solutions Manual for Organic Chemistry Janice Gorzynski Smith, Dr., Erin Smith Berk, 2013-02-05 Written by Janice Gorzynski Smith and Erin Smith Berk the Student Study Guide Solutions Manual provides step by step solutions to all in chapter and end of chapter problems Each chapter begins with an overview of key concepts and includes a short answer practice test on the fundamental principles and new reactions

Biochemistry David E. Metzler, Carol M. Metzler, 2001 Biochemistry The Chemical Reactions of Living Cells is a well integrated up to date reference for basic chemistry and underlying biological phenomena Biochemistry is a comprehensive account of the chemical basis of life describing the amazingly complex structures of the compounds that make up cells the forces that hold them together and the chemical reactions that allow for recognition signaling and movement This book contains information on the human body its genome and the action of muscles eyes and the brain Thousands of literature references provide introduction to current research as well as historical background Contains twice the number of chapters of the first edition Each chapter contains boxes of information on topics of general interest

An Introduction to Chemical Kinetics Michel Soustelle, 2013-02-07 This book is a progressive presentation of kinetics of the chemical reactions It provides complete coverage of the domain of chemical kinetics which is necessary for the various future users in the fields of Chemistry Physical Chemistry Materials Science Chemical Engineering Macromolecular Chemistry and Combustion It will help them to understand the most sophisticated knowledge of their future job area Over 15 chapters this book present the fundamentals of chemical kinetics its relations with reaction mechanisms and kinetic properties Two chapters are then devoted to experimental results and how to calculate the kinetic laws in both homogeneous and heterogeneous systems The following two chapters describe the main approximation modes to calculate

these laws Three chapters are devoted to elementary steps with the various classes the principles used to write them and their modeling using the theory of the activated complex in gas and condensed phases Three chapters are devoted to the particular areas of chemical reactions chain reactions catalysis and the stoichiometric heterogeneous reactions Finally the non steady state processes of combustion and explosion are treated in the final chapter **Holt Chemistry**, 2003-01-24

Chemical Engineering Fluid Mechanics Ron Darby, Raj P. Chhabra, 2016-11-30 This book provides readers with the most current accurate and practical fluid mechanics related applications that the practicing BS level engineer needs today in the chemical and related industries in addition to a fundamental understanding of these applications based upon sound fundamental basic scientific principles The emphasis remains on problem solving and the new edition includes many more examples **An Introduction to Chemistry** Mark Bishop, 2002 This book teaches chemistry at an appropriate level of rigor while removing the confusion and insecurity that impair student success Students are frequently intimidated by prep chem Bishop s text shows them how to break the material down and master it The flexible order of topics allows unit conversions to be covered either early in the course as is traditionally done or later allowing for a much earlier than usual description of elements compounds and chemical reactions The text and superb illustrations provide a solid conceptual framework and address misconceptions The book helps students to develop strategies for working problems in a series of logical steps The Examples and Exercises give plenty of confidence building practice the end of chapter problems test the student s mastery The system of objectives tells the students exactly what they must learn in each chapter and where to find it **Chemistry 2e** Paul Flowers, Klaus Theopold, Richard Langley, Edward J. Neth, William R. Robinson, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition Holt Chemistry R. Thomas Myers, 2006 Solving General Chemistry Problems Robert Nelson Smith, Willis Conway Pierce, 1980-01-01

Chemistry Bruce Averill, Patricia Eldredge, 2007 Emphasises on contemporary applications and an intuitive problem solving approach that helps students discover the exciting potential of chemical science This book incorporates fresh applications from the three major areas of modern research materials environmental chemistry and biological science

Chemistry in Context AMERICAN CHEMICAL SOCIETY., 2024-04-11 **Fundamentals of General, Organic, and Biological Chemistry** John McMurry, 2013 Fundamentals of General Organic and Biological Chemistry by McMurry

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- Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information
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