

Empirical and Molecular Formulas Worksheet 1

1. The percentage composition of acetic acid is found to be 39.9% C, 6.7% H, and 53.4% O. Determine the empirical formula of acetic acid.

$$\frac{39.9\% \text{ C}}{12.01\%} \times \frac{1-1}{3.72} = \frac{3.32-1}{3.72} = 1 \quad \frac{3.4\% \text{ H}}{1.008\%} \times \frac{1-1}{2.32} = \frac{3.34-1}{2.32} = 1$$

$$\frac{6.7\% \text{ H}}{1.008\%} \times \frac{1-1}{3.72} = \frac{6.6-1}{3.72} = 2$$



- a. The molar mass for question #1 was determined by experiment to be 60.0 g/mol. What is the molecular formula?

$$\frac{60.0}{30.03} = 2 \quad \text{C}_2\text{H}_4\text{O}_2$$

2. Aniline, a starting material for urethane plastic foams, consists of C, H, and N. Combustion of such compounds yields CO_2 , H_2O , and N_2 as products. If the combustion of 9.71 g of aniline yields 6.63 g H_2O and 1.46 g N_2 , what is its empirical formula?

$$\frac{6.63\text{g } \text{H}_2\text{O}}{18.02\text{g}} \times \frac{1-1}{2-1} = \frac{.736-1\text{H}}{1.04} = 7 \quad \frac{1.46\text{g } \text{N}_2}{28.02\text{g}} \times \frac{1-1}{2-1} = \frac{1.46-1\text{N}}{1.4} = 1$$

$$\frac{9.71\text{g} - 6.63\text{g} - 1.46\text{g}}{12.01\text{g}} = \frac{1.62\text{g}}{12.01\text{g}} = \frac{.135-1\text{C}}{1.04} = 1$$



- a. The molar mass of aniline is 93 g/mol. What is its molecular formula?

$$\frac{93}{93.13} = 1 \quad \text{C}_6\text{H}_7\text{N}$$

3. Calculate the mass percent of carbon, nitrogen and oxygen in acetamide, $\text{C}_2\text{H}_5\text{NO}$.

$\frac{2(12.01)}{59.07}$	$\frac{5(1.008)}{59.07}$	$\frac{1(14.01)}{59.07}$	$\frac{1(16.00)}{59.07}$
40.66% C	8.52% H	23.72% N	27.04% O

4. A 50.51 g sample of a compound made from phosphorus and chlorine is decomposed. Analysis of the products showed that 11.39 g of phosphorus atoms were produced. What is the empirical formula of the compound?

$$\frac{11.39\text{g P}}{30.97\text{g}} \times \frac{1-1}{.3679} = 1$$

$$50.51 - 11.39 = 39.12\text{g Cl} \quad \frac{39.12\text{g}}{35.45\text{g}} \times \frac{1-1}{.3679} = 3$$



Empirical Formulas Worksheet 1 Answers

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