

5. The ionic product of water,  $K_w$  at 298K =  $1 \times 10^{-14}$ .

(a) Given the  $K_w$  of water at 50°C is  $5.47 \times 10^{-14}$  calculate the pH of water at this temperature. (3)

(b) From your answer to (a) and your knowledge of Le Chatelier's Principle, suggest and explain whether the ionization of water is exothermic or endothermic. (2)

Chemistry Pages

# **Adrian Dingle Chemistry Pages And Chemeducator Llc 2013 Answers**

**Robert R. Redfield**



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