ChemActivity 40

Part A: Reactions of Carboxylic Acids

(How does the acidic O-H group affect nucleophilic addition to the carbonyl carbon?)

Model 1: Acid-Base Reactivity of Carboxylic Acids



Critical Thinking Questions

- How many different kinds of hydrogens are there on acetic acid?
 - Draw both possible conjugate bases INCLUDING ANY IMPORTANT RESONANCE STRUCTURES.
 - Circle the conjugate base that is lower in potential energy and explain your reasoning.
- Methoxide ion ("OCH₃) is an excellent nucleophile, yet the following nucleophilic addition does not occur.



- a) Which is higher in energy: the nucleophilic addition product above or the conjugate base of acetic acid? [Cross out the one that is higher in energy.]
- b) Is your answer in part a) consistent with the fact that with methoxide the acid base reaction is favored over the nucleophilic addition reaction? Explain.

Chemactivity 40 Answers

Michael P. Garoutte

Chemactivity 40 Answers

Organic Chemistry, a Guided Inquiry Andrei Straumanis, 2004 Chemistry 2e Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition **Chemistry** Richard S. Moog, John J. Farrell, 2017-06-26 In the newly updated 7th Edition Chemistry A Guided Inquiry continues to follow the underlying principles developed by years of extensive research on how students learn and draws on testing by those using the POGIL methodology This text follows the principles of inquiry based learning and correspondingly emphasizes underlying chemistry concepts and the reasoning behind them This text provides an approach that follows modern cognitive learning principles by having students learn how to create knowledge based on experimental data and how to test that knowledge Pearson Chemistry Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, 2012-01-01 **Holt McDougal Modern Chemistry** Mickey Sarguis, 2012 Chemistry Steven S. Zumdahl, Susan A. Zumdahl, 2007 Contains discussion illustrations and exercises aimed at overcoming common misconceptions emphasizes on models prevails and covers topics such as chemical foundations types of chemical reactions and solution stoichiometry electrochemistry and organic and biological molecules Organic Chemistry David R. Klein, 2017-08-14 In Organic Chemistry 3rd Edition Dr David Klein builds on the phenomenal

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Activity Coefficients in Electrolyte Solutions Kenneth S.

Pitzer, 2018-05-04 This book was first published in 1991 It considers the concepts and theories relating to mostly aqueous systems of activity coefficients

General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to

a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration AP Chemistry For Dummies Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out or your AP course You ll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You ll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way around laboratory concepts tasks equipment and safety Analyze laboratory data Use practice exams to maximize your score Additionally you ll have a chance to brush up on the math skills that will help you on the exam learn the critical types of chemistry problems and become familiar with the annoying exceptions to chemistry rules Get your own copy of AP Chemistry For Dummies to build your confidence and test taking know how so you can ace that exam Green Toxicoloav Alexandra Maertens, 2022-02-01 Green toxicology is an integral part of green chemistry. One of the key goals of green chemistry is to design less toxic chemicals Therefore an understanding of toxicology and hazard assessment is important for any chemist working in green chemistry but toxicology is rarely part of most chemists education As a consequence chemists lack the toxicological lens necessary to view chemicals in order to design safer substitutions This book seeks to fill that gap and demonstrate how a basic understanding of toxicology as well as the tools of in silico and in vitro toxicology can be an integral part of green chemistry R D chemists product stewards and toxicologists who work in the field of sustainability can all benefit from integrating green toxicology principles into their work Topics include in silico tools for hazard assessment toxicity testing and lifecycle considerations this book aims to act as a bridge between green toxicologists and green chemists

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according to which the active site on a catalyst surface has a supra molecular character This the first book on the subject is the outcome of a NATO Workshop held in Le Rouret France in May 1986 It is our hope that the following chapters and the concluding summary of recommendations for research may help to provide a definition of surface organometallic chemistry Besides catalysis the central theme of the Workshop four main topics are considered 1 Reactions of organometallics with surfaces of metal oxides metals and zeolites 2 Molecular models of surfaces metal oxides and metals 3 Molecular approaches to the mechanisms of surface reactions 4 Synthesis and modification of zeolites and related microporous solids Most surface organometallic chemistry has been carried out on amorphous high surface area metal oxides such as silica alumina magnesia and titania The first chapter contributed by KNOZINGER gives a short summary of the structure and reactivity of metal oxide surfaces Most of our understanding of these surfaces is based on acid base and redox chemistry this chemistry has developed from X ray and spectroscopic data and much has been inferred from the structures and reactivities of adsorbed organic probe molecules There are major opportunities for extending this understanding by use of well defined single crystal oxide surfaces and organometallic probe molecules Merrill Chemistry Robert C. Smoot, Smoot, Richard G. General Chemistry Ralph H. Petrucci, F. Geoffrey Herring, Jeffry D. Madura, Carey Smith, Jack Price, 1998 Bissonnette, 2010-05 Specific Ion Effects Werner Kunz, 2010 Specific ion effects are important in numerous fields of science and technology This book summarizes the main ideas that came up over the years It presents the efforts of theoreticians and supports it by the experimental results stemming from various techniques Basic Principles of Organic Chemistry John D. Roberts, Marjorie C. Caserio, 1977 Introduction what is organic chemistry all about Structural organic chemistry the shapes of molecules functional groups Organic nomenclature Alkanes Stereoisomerism of organic molecules Bonding in organic molecules atomic orbital models More on nomenclature compounds other than hydrocarbons Nucleophilic substitution and elimination reactions Separation and purification identification of organic compounds by spectroscopic techniques Alkenes and alkynes Ionic and radical addition reactions Alkenes and alkynes Oxidation and reduction reactions Acidity or alkynes

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