

DAT General Chemistry Equation Sheet

General Concepts

Density: $(d = m/v)$

Dilutions: $M_1V_1 = M_2V_2$ (or $C_1V_1 = C_2V_2$)

% Error Calculations

$$\% \text{ Error} = \frac{\text{Actual} - \text{theoretical}}{\text{theoretical}} \times 100$$

Spectrophotometer: $Abs = \epsilon cl$

Abs = Absorbance

ϵ = molar extinction coefficient (molar absorptivity)

l = path length

Atomic Structure

Absorption/Emission Line Spectra: $\Delta E = E_{\text{photon}}$

Energy of a photon: $E_{\text{photon}} = hf = \frac{hc}{\lambda}$

$c = 3.0 \times 10^8 \text{ m/s}$

$h = 6.63 \times 10^{-34} \text{ J} \cdot \text{s}$

Photoelectric Effect: $KE_{\text{electron}} = E_{\text{photon}} - \phi$

ϕ = work function (minimum energy to ionize electron)

Chemical Kinetics and Equilibrium

General Rate Law: $\text{rate} = k[A]^m[B]^n$

Arrhenius Equation: $k = Ae^{-E_a/RT}$ E_a = activation energy

$R = 0.0821 \text{ L atm/mol K}$

Equilibrium Constant:

$$K_c = \frac{[\text{products}]}{[\text{reactants}]} \quad K_p = \frac{P_{\text{products}}}{P_{\text{reactants}}}$$

Solutions

Concentration

$$\text{Molarity (M)} = \frac{\text{moles}_{\text{solute}}}{L_{\text{solution}}}$$

$$\text{molality (m)} = \frac{\text{moles}_{\text{solute}}}{kg_{\text{solvent}}}$$

Henry's Law: $P_A = k_H[A]$

Colligative Properties

Freezing Point Depression:

$$\Delta T_f = -iK_f m$$

Boiling Point Elevation:

$$\Delta T_b = iK_b m$$

Raoult's Law (* = pure):

$$P_A = X_A P_A^*$$

Osmotic Pressure:

$$\pi = iMRT$$

K_f/K_b = F.P./B.P. depression/elevation constant

X_A = mole fraction of A

i = van't Hoff Factor

$R = 0.0821 \text{ L atm/mol K}$

M = molarity, m = molality

Gases/Gas Laws

Pressure: $P = \text{Force/Area}$

Average Kinetic Energy:

$K.E._{avg} = \frac{3}{2} RT$ (per mole of gas)

Root-mean-square speed: $v = \sqrt{\frac{3RT}{M_m}}$

Gas Laws

Ideal Gas Law:

$$PV = nRT$$

Boyle's Law:

$$P \propto \frac{1}{V}$$

Charles' Law:

$$V \propto T$$

Avogadro's Law:

$$V \propto n$$

Combined Gas Law:

$$\frac{P_1 V_1}{n_1 T_1} = \frac{P_2 V_2}{n_2 T_2}$$

Dalton's Law of Partial Pressures:

$$P_{\text{Tot}} = P_A + P_B + \dots$$

$$P_A = X_A \cdot P_{\text{Tot}}$$

$$X_A = \text{mol fraction } A$$

$R = 0.0821 \text{ L atm/mol K}$

n = moles

M_m = molar mass

r = rate of effusion

a and b are constants specific to a given gas.

Grahams Law of Effusion:

$$\frac{r_1}{r_2} = \sqrt{\frac{M_{m2}}{M_{m1}}}$$

Acid-Base

pH Scale

$$pH = -\log[H^+]$$

$$[H^+] = 10^{-pH}$$

$$pOH = -\log[OH^-]$$

$$[OH^-] = 10^{-pOH}$$

$$pH + pOH = 14$$

$$[H^+][OH^-] = 1 \times 10^{-14}$$

Weak Acids:

$$K_a = \frac{[H_3O^+][A^-]}{[HA]}$$

$$[H^+] = \sqrt{K_a[HA]}$$

Weak Bases:

$$K_b = \frac{[OH^-][HA]}{[A^-]}$$

$$[OH^-] = \sqrt{K_b[B]}$$

$$K_a \cdot K_b = K_w$$

Buffer: $pH = pK_a + \log \frac{[A^-]}{[HA]}$

Titration/Neutralization: $n_a M_a V_a = n_b M_b V_b$

Chemical Engineering Formula Sheet

R Barnett



Chemical Engineering Formula Sheet

Formulas, Facts, and Constants H. J. Fischbeck, K. H. Fischbeck, 2012-12-06 The material presented in this book has been compiled for the convenience of the reader The aim of the book is to provide a handy source of formulas conversion factors and constants for everyday use The formulas and tables are amended by examples in all of those cases where their use is not self explanatory The material has been selected to be helpful whenever it is inconvenient or not possible to consult tables available at the library Section 1 provides the fundamental tools of mathematics needed in all areas of the physical sciences Section 2 summarizes the SI system lists conversion factors and provides precise values of fundamental constants Sections 3 and 4 review the basic terms of spectroscopy atomic structure and wave mechanics These sections serve as a guide to the interpretation of modern literature Section 5 is a resource for work in the laboratory Data and formulas are given to be of assistance in the use of frequently encountered equipment such as vacuum systems and electronic devices Material constants and other data are listed for information and as an aid for estimates or problem solving The assistance of the Springer Verlag during the various stages of the development of this book is gratefully acknowledged The authors like to thank Dr F L Boschke for his many helpful suggestions Helmut J Fischbeck Kurt H Fischbeck Contents Basic mathematical facts and figures *Handbook of Chemical Engineering Calculations* Nicholas P. Chopey, 1994 A compilation of the calculation procedures needed every day on the job by chemical engineers Tables of Contents Physical and Chemical Properties Stoichiometry Phase Equilibrium Chemical Reaction Equilibrium Reaction Kinetics and Reactor Design Flow of Fluids and Solids Heat Transfer Distillation Extraction and Leaching Crystallization Filtration Liquid Agitation Size Reduction Drying Evaporation Environmental Engineering in the Plant Illustrations Index *Chemical Engineering Design* Gavin Towler, Ray Sinnott, 2012-01-25 Chemical Engineering Design Second Edition deals with the application of chemical engineering principles to the design of chemical processes and equipment Revised throughout this edition has been specifically developed for the U S market It provides the latest US codes and standards including API ASME and ISA design codes and ANSI standards It contains new discussions of conceptual plant design flowsheet development and revamp design extended coverage of capital cost estimation process costing and economics and new chapters on equipment selection reactor design and solids handling processes A rigorous pedagogy assists learning with detailed worked examples end of chapter exercises plus supporting data and Excel spreadsheet calculations plus over 150 Patent References for downloading from the companion website Extensive instructor resources including 1170 lecture slides and a fully worked solutions manual are available to adopting instructors This text is designed for chemical and biochemical engineering students senior undergraduate year plus appropriate for capstone design courses where taken plus graduates and lecturers tutors and professionals in industry chemical process biochemical pharmaceutical petrochemical sectors New to this edition Revised

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Formulas, Facts, and Constants for Students and Professionals in Engineering, Chemistry, and Physics Helmut J. Fischbeck, Kurt Fischbeck, 1982 *Short Handbook of Mathematical Formulas for Chemical Engineers* Muhammad Rashid Usman, 2017-09-20 This book is a collection of physical constants conversion factors mathematical formulas and some essential MATLAB commands many of which are required on daily basis by chemical engineers The book is intended to save vital time of the students instructors researchers and people in the process industry which they have to spend in searching the material available in this book

Rules of Thumb for Chemical Engineers Carl Branan, 2002 Fractionators separators and accumulators cooling towers gas treating blending troubleshooting field cases gas solubility and density of irregular solids Hundreds of common sense techniques shortcuts and calculations

Chemical Engineering Calculations Ernest J. Henley, Herman Bieber, 1959

Chemical Engineering Fluid Mechanics Ron Darby, Raj P. Chhabra, 2016-11-30 This book provides readers with the most current accurate and practical fluid mechanics related applications that the practicing BS level engineer needs today in the chemical and related industries in addition to a fundamental understanding of these applications based upon sound fundamental basic scientific principles The emphasis remains on problem solving and the new edition includes many more examples

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Henley's Formulas, Recipes and Processes Various, 2023-12-12 Henley's Formulas Recipes and Processes is a comprehensive reference book that compiles a vast array of formulas recipes and processes for use in a variety of industries such as chemistry engineering and manufacturing The book is written in a straightforward and practical style making it accessible to professionals and students alike Each formula is meticulously explained and organized providing valuable insights into the practical application of scientific principles This text stands out as a valuable resource for anyone seeking to expand their knowledge in the field of applied sciences The extensive content covers a wide range of topics making it a must have for anyone working in these fields Various the author of Henley's Formulas Recipes and Processes is a collective pseudonym used by a group of experts in various scientific disciplines Their combined knowledge and experience have enabled them to compile this extensive and authoritative reference book This collaborative effort showcases the diversity and depth of expertise within the scientific community I highly recommend Henley's Formulas Recipes and Processes to professionals students and individuals interested in expanding their knowledge of formulas and processes in the fields of science and engineering This book serves as an invaluable tool for anyone seeking practical and in depth information on a wide range of topics

The AIChE Pocket Handbook American Institute of Chemical Engineers, 1985 Chemical Engineering Catalog, 1919 Chemistry 2e Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition

The Chemist's Handbook, 1922 **Elements of Chemical Reaction Engineering** H. Scott Fogler, 1999 The fourth edition of Elements of Chemical Reaction Engineering is a completely revised version of the book It combines authoritative coverage of the principles of chemical reaction engineering with an unsurpassed focus on critical thinking and creative problem solving employing open ended questions and stressing the Socratic method Clear and organized it integrates text visuals and computer simulations to help readers solve even the most challenging problems through reasoning rather than by memorizing equations BOOK JACKET *Basic Principles and Calculations in Chemical*

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Analysis, Synthesis and Design of Chemical Processes Richard Turton, Richard C. Bailie, Wallace B. Whiting, Joseph A. Shaeiwitz, 2008-12-24 The Leading Integrated Chemical Process Design Guide Now with New Problems New Projects and More More than ever effective design is the focal point of sound chemical engineering Analysis Synthesis and Design of Chemical Processes Third Edition presents design as a creative process that integrates both the big picture and the small details and knows which to stress when and why Realistic from start to finish this book moves readers beyond classroom exercises into open ended real world process problem solving The authors introduce integrated techniques for every facet of the discipline from finance to operations new plant design to existing process optimization This fully updated Third Edition presents entirely new problems at the end of every chapter It also adds extensive coverage of batch process design including realistic examples of equipment sizing for batch sequencing batch scheduling for multi product plants improving production via intermediate storage and parallel equipment and new optimization techniques specifically for batch processes Coverage includes Conceptualizing and analyzing chemical processes flow diagrams tracing process conditions and more Chemical process economics analyzing capital and manufacturing costs and predicting or assessing profitability Synthesizing and optimizing chemical processing experience based principles BFD PFD simulations and more Analyzing process performance via I O models performance curves and other tools Process troubleshooting and debottlenecking Chemical engineering design and society ethics professionalism health safety and new green engineering techniques Participating successfully in chemical engineering design teams Analysis Synthesis and Design of Chemical Processes Third Edition draws on nearly 35 years of innovative chemical engineering instruction at West Virginia University It includes suggested curricula for both single semester and year long design courses case studies and design projects with practical applications and appendixes

with current equipment cost data and preliminary design information for eleven chemical processes including seven brand new to this edition *Physico-chemical Tables for the Use of Analysts, Physicists, Chemical Manufacturers and Scientific Chemists* John Castell-Evans,1902

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