

ENGAGE

Why does the atomic number but not the mass number change in positron emission?

SAMPLE PROBLEM 5.4 Writing a Nuclear Equation for Positron Emission**TRY IT FIRST**

Write the balanced nuclear equation for manganese-49, which decays by emitting a positron.

SOLUTION GUIDE

	Given	Need	Connect
ANALYZE THE PROBLEM	Mn-49, positron emission	balanced nuclear equation	mass number, atomic number of new nucleus

STEP 1 Write the incomplete nuclear equation.



STEP 2 Determine the missing mass number. In the equation, the mass number, 49, is equal to the sum of the mass numbers of the new nucleus and the positron.

$$49 = ? + 0$$

$$49 - 0 = ?$$

$$49 - 0 = 49 \text{ (mass number of new nucleus)}$$

STEP 3 Determine the missing atomic number. The atomic number, 25, must equal the sum of the atomic numbers of the new nucleus and the positron.

$$25 = ? + 1$$

$$25 - 1 = ?$$

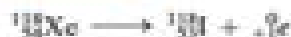
$$25 - 1 = 24 \text{ (atomic number of new nucleus)}$$

STEP 4 Determine the symbol of the new nucleus. On the periodic table, the element that has atomic number 24 is chromium, Cr. The atomic symbol for this isotope of Cr is written ${}_{24}^{49}\text{Cr}$.

STEP 5 Complete the nuclear equation.

**STUDY CHECK 5.4**

Write the balanced nuclear equation for xenon-118, which undergoes positron emission.

ANSWER**TEST**

Try Practice Problems 5.15 to 5.18

Pearson Mastering Chemistry Answers

Laura D. Frost, S. Todd Deal



Pearson Mastering Chemistry Answers

Pearson Chemistry Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, 2012-01-01

Chemistry OpenStax, 2014-10-02 This is part one of two for Chemistry by OpenStax This book covers chapters 1-11

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General, Organic, and Biological Chemistry Laura D. Frost, S. Todd Deal, 2016-01-20 A Concise Introduction to General Organic and Biological Chemistry General Organic and Biological Chemistry strengthens the evidenced strategy of integrating general organic and biological chemistry for a focused introduction to the fundamental connections between chemistry and life The streamlined approach offers readers a clear path through the content over a single semester The Third Edition integrates essential topics more effectively than any text on the market covering core concepts in each discipline in just 12 comprehensive chapters Practical connections and applications show readers how to use their understanding of chemistry in everyday life and future health professions With an emphasis on problem solving and critical thinking the book promotes active and attentive learning which now include NEW media assets Practicing the Concepts Featuring coauthor Todd Deal these 3 to 5 minute videos explore key concepts in general organic and biological chemistry that readers traditionally find difficult Readers gain skills and deepen their knowledge as they watch the videos and then practice what they have learned with Pause Pearson eText and Mastering Chemistry do not come packaged with this content Students ask your instructor for the correct package ISBN and Course ID Instructors contact your Pearson representative for more information If your instructor has assigned Pearson eText as your main course material search for 0135237327 9780135237328 Pearson eText General Organic and Biological Chemistry 3 e Access Card OR 0135237335 9780135237335 Pearson eText General Organic and Biological Chemistry 3 e Instant Access If you would like to purchase both the physical text and MasteringChemistry search for 0134041569 9780134041568 General Organic and Biological Chemistry Plus MasteringChemistry with eText Access Card Package 3 e Package consists of 0134162048 9780134162041

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