

- (ii) How many elements are in the second period?
- (iii) Name the element which has the highest electron affinity.
- (iv) Name the element which has the highest electro negativity.
- (v) Name the element which may be placed on group 1 but is not a metal. [5]

(b) Fill in the blanks using the correct options:

- (i) Metals have ----- ionisation potential. (low/ high)
- (ii) Group 18 elements have ----- valence electrons (4 / 8) with the exception of ---- (He / Ne) with ----- electrons (2 / 8) in valence shell.
- (iii) Group 2 elements are called ----- metals (alkali / alkaline earth). [5]

Question 3.

(a) Draw different isomers having the following molecular formula:

- (i) C_5H_{12} (chain)
- (ii) C_4H_8 (position). [5]

(b) What is denatured alcohol? [1]

(c) Give two important uses of ethanol. [2]

(d) Write equations for:

- (i) Preparation of ethanol by hydration of C_2H_4 .
- (ii) Preparation of acetic acid from ethanol. [2]

Question. 4

(a) Name the method by which following compounds can be prepared:

Select the appropriate method from the following list Neutralization; direct combination; precipitation; metal + acid – use a method only once.

- (i) Sodium sulphate
- (ii) Silver chloride
- (iii) Iron sulphide. [3]

(b) How will you distinguish between following pairs of compounds using NH_4OH .

- (i) Copper sulphate and iron(II) sulphate.

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