

# THE MOLE AND AVOGADRO'S NUMBER

Name \_\_\_\_\_

(atoms)

One mole of a substance contains Avogadro's Number ( $6.02 \times 10^{23}$ ) of molecules.

How many molecules are in the quantities below?

1. 2.0 moles  $x = 2.0 \text{ mol} \times \left( \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \right) = \underline{1.2 \times 10^{24} \text{ molecules}}$

2. 1.5 moles  $x = 1.5 \text{ mol} \times \left( \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \right) = \underline{9.0 \times 10^{23} \text{ molecules}}$

3. 0.75 mole  $x = 0.75 \text{ mol} \times \left( \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \right) = \underline{4.5 \times 10^{23} \text{ molecules}}$

4. 15 moles  $x = 15 \text{ mol} \times \left( \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \right) = \underline{9.0 \times 10^{24} \text{ molecules}}$

5. 0.35 mole  $x = 0.35 \text{ mol} \times \left( \frac{6.02 \times 10^{23} \text{ molecules}}{1 \text{ mol}} \right) = \underline{2.1 \times 10^{23} \text{ molecules}}$

How many moles are in the number of molecules below?

1.  $6.02 \times 10^{23}$   $x \text{ moles} = 6.02 \times 10^{23} \text{ molecules} \left( \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ molecules}} \right) = \underline{1.0 \text{ mol}}$

2.  $1.204 \times 10^{24}$   $x \text{ moles} = 1.204 \times 10^{24} \text{ molecules} \left( \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ molecules}} \right) = \underline{2.0 \text{ moles}}$

3.  $1.5 \times 10^{20}$   $x \text{ moles} = 1.5 \times 10^{20} \text{ molecules} \left( \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ molecules}} \right) = \underline{2.5 \times 10^{-4} \text{ moles}}$  (0.00024 moles)

4.  $3.4 \times 10^{26}$   $x \text{ moles} = 3.4 \times 10^{26} \text{ molecules} \left( \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ molecules}} \right) = \underline{565 \text{ moles}}$

5.  $7.5 \times 10^{19}$   $x \text{ moles} = 7.5 \times 10^{19} \text{ molecules} \left( \frac{1 \text{ mol}}{6.02 \times 10^{23} \text{ molecules}} \right) = \underline{1.2 \times 10^{-4} \text{ moles}}$

# **Instructional Fair Chemistry If8766**

**Harley Laroux**



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*Chemistry Homework* Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more Chemistry, Grades 9 - 12 Joan Distasio, 1999-01-15 Activity sheets to enhance chemistry lessons at any level Includes problems and puzzles on the mole balancing equations gas laws stoichiometry and the periodic table OCLC Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Aircraft Instrumentation and Systems S.** Nagabhushana, 2013-12-30 Aircraft Instrumentation and Systems has the adequate coverage to deal generally the topics for undergraduate course on Aircraft Instrumentation It covers An introduction to aircraft instruments and systems Air data systems and air data computers Navigation systems Gyroscopic flight instruments Engine instruments Electronics flight instrument systems Safety and warning systems Every effort has been done to update the contents of the book to the present day technology used in modern transport category aircraft manufactured by Boeing and Airbus industry The text is profusely illustrated with block diagrams schematic diagrams and a number of tables and glossary Review questions have been included at the end of the each chapter for practice and self study The book is intended for teaching and study the topic for students of B E M E and students in Instrumentation Technology and Aircraft Engineering It also introduces the subject to practising engineers and readers interested in aircraft instrumentation and to the flight crew **Fundamentals of General, Organic, and Biological Chemistry** John McMurry, 2013 Fundamentals of General Organic and Biological Chemistry by McMurry Ballantine Hoeger and Peterson provides background in chemistry and biochemistry with a relatable context to ensure students of all disciplines gain an appreciation of chemistry s significance in everyday life Known for its clarity and concise presentation this book balances chemical concepts with examples drawn from students everyday lives and experiences to explain the quantitative aspects of chemistry and provide deeper insight into theoretical principles The Seventh Edition focuses on making connections between General Organic and Biological Chemistry through a number of new and updated features including all new Mastering Reactions boxes Chemistry in Action boxes new and revised chapter problems that strengthen the ties between major concepts in each chapter practical applications and much more NOTE this

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Biological Chemistry **Who's Hiding?** Satoru Onishi,2009-08 The reader is asked a question about each page of animal  
pictures **Lakeland:** Lakeland Community Heritage Project Inc.,2012-09-18 Lakeland the historical African American  
community of College Park was formed around 1890 on the doorstep of the Maryland Agricultural College now the University  
of Maryland in northern Prince George s County Located less than 10 miles from Washington D C the community began when  
the area was largely rural and overwhelmingly populated by European Americans Lakeland is one of several small African  
American communities along the U S Route 1 corridor between Washington D C and Laurel Maryland With Lakeland s  
central geographic location and easy access to train and trolley transportation it became a natural gathering place for  
African American social and recreational activities and it thrived until its self contained uniqueness was undermined by the  
federal government s urban renewal program and by societal change The story of Lakeland is the tale of a community that  
was established and flourished in a segregated society and developed its own institutions and traditions including the area s  
only high school for African Americans built in 1928 Children's Books in Print, 2007 ,2006 *Chemistry & Chemical  
Reactivity* John C. Kotz,Paul Treichel,1999 The principal theme of this book is to provide a broad overview of the principles of  
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Secrets of Alchemy* Lawrence Principe,2013 Alchemy the Noble Art conjures up scenes of mysterious dimly lit laboratories  
populated with bearded old men stirring cauldrons Though the history of alchemy is intricately linked to the history of  
chemistry alchemy has nonetheless often been dismissed as the realm of myth and magic or fraud and pseudoscience And  
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## **Instructional Fair Chemistry If8766** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the energy of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such may be the essence of the book **Instructional Fair Chemistry If8766**, a literary masterpiece that delves deep to the significance of words and their affect our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall impact on readers.

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