

Question: What happens to particles when heated?

TEMPERATURE AND HEAT

Temperature – a measure of the average value of the kinetic energy of the molecules in random motion. (SI Unit for temperature is Kelvin (K)).

Thermal expansion – almost all substances expand when they are Heated and contract when they are cooled ----- exception water

Thermal energy – sum of the kinetic and potential energy of all the particles in an object; thermal energy of an object increases as temperature increases

Temperature Conversion Equations	
$^{\circ}\text{F} \rightarrow ^{\circ}\text{C}$	$^{\circ}\text{C} \rightarrow ^{\circ}\text{F}$
$^{\circ}\text{C} = (5/9)(^{\circ}\text{F} - 32)$	$^{\circ}\text{F} = (9/5)^{\circ}\text{C} + 32$

Heat – thermal energy that flows from something at a higher temperature to something at a lower temp.

Specific heat – amt of heat Needed to raise the temp of 1 kg of some material by 1°C

Specific Heat of Common Materials	
Substance	Spec. Heat (J/(kg°C))
Water	4,184
Wood	1,760
Carbon	710
Glass	664
Iron	450

Thermal Energy Equation

(Q) Change in thermal energy (J) =
Mass (kg) x Δ temp (°C) x
(c) Specific heat (J/kg°C)

$$Q = m(T_f - T_i) c$$

Assessment Chapter Test Thermal Energy Physical Science

M Mosston



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Simplified Physics Calculations

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