

Thermodynamic Chemistry – Class Level Multiple Choice

Directions: Please show your work on this page.

Name:

<p>1. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant pressure. The amount of heat added is Q. What is the change in internal energy of the substance?</p>	<p>2. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant volume. The amount of heat added is Q. What is the change in internal energy of the substance?</p>	<p>3. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant pressure. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>
<p>4. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant volume. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>	<p>5. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant pressure. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>	<p>6. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant volume. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>
<p>7. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant pressure. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>	<p>8. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant volume. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>	<p>9. A substance of mass m and specific heat capacity c is heated from T_1 to T_2 at constant pressure. The amount of heat added is Q. What is the change in enthalpy of the substance?</p>

10. $Q = mc(T_2 - T_1)$

11. $Q = mc(T_2 - T_1) + p\Delta V$

12. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H$

13. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H + \Delta G$

14. $Q = mc(T_2 - T_1) + p\Delta V$

15. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H$

16. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H + \Delta G$

17. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H + \Delta G + \Delta A$

18. $Q = mc(T_2 - T_1)$

19. $Q = mc(T_2 - T_1) + p\Delta V$

20. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H$

21. $Q = mc(T_2 - T_1) + p\Delta V + \Delta H + \Delta G$

Technical Chemistry Gas Laws Magic Square Answer Key

David Baud



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