

CHEMISTRY: LIMITING REAGENT Practice Problems

Please show your work on a separate piece of paper!

Assume standard temperature and pressure, STP conditions for all reactions.

1. At high temperatures, sulfur combines with iron to form the brown-black iron (II) sulfide:



In one experiment, 7.62 g of Fe are allowed to react with 8.67 g of S.

- What is the limiting reagent, and what is the reactant in excess?
 - Calculate the mass of FeS formed.
2. Acrylonitrile, $\text{C}_3\text{H}_3\text{N}$, is the starting material for the production of a kind of synthetic fiber (acrylics) and can be made from propylene, C_3H_6 , by reaction with nitric oxide, NO , as follows:



What mass of $\text{C}_3\text{H}_3\text{N}$ can be made when 21.6 g of C_3H_6 react with 21.6 g of nitric oxide?

3. Calculate the percent yield for the reaction: $\text{P}_4_{(s)} + 6 \text{Cl}_2_{(g)} \rightarrow 4 \text{PCl}_3_{(l)}$
if 75.0 g of phosphorus reacts with excess chlorine gas to produce 111.0 g of phosphorus trichloride.
4. Formic acid, HCHO_2 , burns in oxygen to form carbon dioxide and water as follows:



If a 3.15-g sample of formic acid was burned in 2.0 L of oxygen, what volume of carbon dioxide would be produced?

5. Zinc metal reacts with hydrochloric acid to produce zinc chloride and hydrogen gas.
- Balance the following reaction: $\text{Zn}_{(s)} + \text{HCl}_{(aq)} \rightarrow \text{ZnCl}_2_{(aq)} + \text{H}_2_{(g)}$
 - A 3.50-g sample of zinc metal is allowed to react with 2.50 g of hydrochloric acid.

Complete the following table:

reactants/products	Zn (grams)	HCl (grams)	ZnCl ₂ (grams)	H ₂ (L)
Before reaction				
After reaction	1.26 g			

Limiting Reagent Problems Answers

**Steven S. Zumdahl, Susan L.
Zumdahl, Donald J. DeCoste**



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Green Organic Chemistry Kenneth M. Doxsee, James E. Hutchison, 2004 This lab text describes the tools and strategies of green chemistry and the lab experiments that allow investigation of organic chemistry concepts and techniques in a greener laboratory setting Students acquire the tools to assess the health and environmental impacts of chemical processes and the strategies to improve develop new processes that are less harmful to human health and the environment The curriculum introduces a number of state of the art experiments and reduces reliance on expensive environmental controls such as fume hoods Provided by publisher

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields

U Can: Chemistry I For Dummies John T. Moore, Chris Hren, Peter J. Mikulecky, 2015-07-21 Now you can score higher in chemistry Every high school requires a course in chemistry for graduation and many universities require the course for majors in medicine engineering biology and various other sciences U Can Chemistry I For Dummies offers all the how to content you need to enhance your classroom learning simplify complicated topics and deepen your understanding of often intimidating course material Plus you'll find easy to follow examples and hundreds of practice problems as well as access to 1 001 additional Chemistry I practice problems online As more and more students enroll in chemistry courses the need for a trusted and accessible resource to aid in study has never been greater That's where U Can Chemistry I For Dummies comes in If you're struggling in the classroom this hands on friendly guide makes it easy to conquer chemistry Simplifies basic chemistry principles Clearly explains the concepts of matter and energy atoms and molecules and acids and bases Helps you tackle problems you may face in your Chemistry I course Combines how to with try it to form one perfect resource for chemistry students If you're confused by chemistry and want to increase your chances of scoring your very best at exam time U Can Chemistry I For Dummies shows you that you can

Dr. Salm's Chemistry Problem Drill Book Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests

General Chemistry I as a Second Language David R. Klein, 2005-03-16 Get a better grade in General Chemistry Even though General Chemistry may be challenging at times with hard work and the right study tools you can still get the grade you want With David Klein's General Chemistry as a Second Language you'll be able to better understand fundamental principles of chemistry solve problems and focus on what you need to know to succeed Here's how you can get a better grade in General Chemistry Understand the basic concepts General Chemistry as a Second Language focuses on selected topics in General Chemistry to give you a solid foundation By understanding these principles you'll have a coherent framework that will help you better understand your course Study more efficiently and effectively General Chemistry as a Second Language provides time saving study tips and problem solving strategies that will help you succeed in the course Improve your problem solving skills General Chemistry as a Second Language will help you develop the skills you need to solve a variety of problem types even unfamiliar ones

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General Chemistry Ralph H. Petrucci, F. Geoffrey Herring, Jeffrey D. Madura, Carey Bissonnette, 2010-05

Solved Problems in Chemistry Beatriz Elena Soledad Rodríguez, 2010-10-26 This book is written to help you study general chemistry. In each subject area, you will find a brief theoretical explanation of the concepts to study, followed by a series of exercises. As you progress through the chapters, you will encounter increasingly complex problems that will be useful for understanding the issues. A number of problems are solved once you understand the explanation, and you are encouraged to try to solve them without seeing the answer to see if you fully understand. In addition to the problems solved, you will be given a series of exercises for you to practice and find the answer to every problem. It is convenient to solve the exercises in the order given to help in reasoning.

Bite-Sized Chemistry Calculations John Lambert, 2024-07-10 *Bite-Sized Chemistry Calculations* is a series of books on chemistry calculations aimed at helping students overcome the challenges associated with tackling the various types of calculations encountered in different aspects of chemistry, focusing on a few topics at a time to facilitate comprehension. Written by an experienced chemistry educator, each book in the series has been tailored to fully meet the needs of students at all levels, especially those taking college-level general chemistry courses as well as those following various O-level curricula worldwide. This part of the series explores the different types of problems and calculations encountered in mass, the mole, and stoichiometry, including the determination of formulae of ionic compounds, relative formula masses, mass and percent compositions of compounds, all aspects of mole calculations, empirical and molecular formulae calculations based on chemical equations, limiting reagents, gas stoichiometry, and percent yield. The series is packed with many salient features that are meant to facilitate both teaching and learning. Some of these include helpful explanations, many examples, alternative ways to solve problems, plenty of practice questions, complete answers, and appendices. With this book, you will be well prepared for your exams and boost your performance.

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Organic Chemistry David R. Klein, 2017-08-14 In *Organic Chemistry*, 3rd Edition, Dr. David Klein builds on the phenomenal success of the first two editions, which presented his unique skills-based approach to learning organic chemistry. Dr. Klein's skills-based approach includes all of the concepts typically covered in an organic chemistry textbook and places special emphasis on skills development to support these concepts. This emphasis on skills development in unique SkillBuilder examples provides extensive opportunities for two-semester Organic Chemistry students to develop proficiency in the key skills necessary to succeed in organic chemistry.

Operational Organic Chemistry John W. Lehman, 1988

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Oswaal Editorial Board, 2023-10-28 Description of the product: Chapter-wise and Topic-wise presentation, Chapter-wise Objectives, A sneak peek into the chapter, Mind Map, A single-page snapshot of the entire chapter, Revision Notes, Concept-based study materials, Tips and Tricks, Useful guidelines for attempting each question perfectly, Some Commonly Made Errors, Most common and unidentified errors are focused, Expert Advice, Oswaal Expert Advice on how to score more, Oswaal QR Codes For Quick Revision on your Mobile Phones and Tablets.

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STOICHIOMETRY AND PROCESS CALCULATIONS K. V. NARAYANAN,B. LAKSHMIKUTTY,2006-01-01 This textbook is designed for undergraduate courses in chemical engineering and related disciplines such as biotechnology polymer technology petrochemical engineering electrochemical engineering environmental engineering safety engineering and industrial chemistry The chief objective of this text is to prepare students to make analysis of chemical processes through calculations and also to develop in them systematic problem solving skills The students are introduced not only to the application of law of combining proportions to chemical reactions as the word stoichiometry implies but also to formulating and solving material and energy balances in processes with and without chemical reactions The book presents the fundamentals of chemical engineering operations and processes in an accessible style to help the students gain a thorough understanding of chemical process calculations It also covers in detail the background materials such as units and conversions dimensional analysis and dimensionless groups property estimation P V T behaviour of fluids vapour pressure and phase equilibrium relationships humidity and saturation With the help of examples the book explains the construction and use of reference substance plots equilibrium diagrams psychrometric charts steam tables and enthalpy composition diagrams It also elaborates on thermophysics and thermochemistry to acquaint the students with the thermodynamic principles of energy balance calculations Key Features SI units are used throughout the book Presents a thorough introduction to basic chemical engineering principles Provides many worked out examples and exercise problems with answers Objective type questions included at the end of the book serve as useful review material and also assist the students in preparing for competitive examinations such as GATE

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