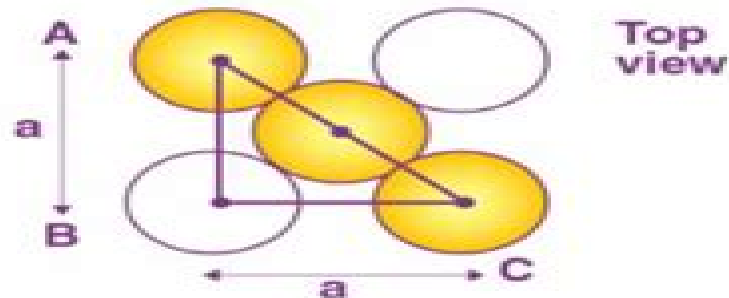


Let the edge length of the unit cell = a
 let the radius of each sphere = r
 Thus, $AC = 4r$
 From the right angled triangle ABC , we have:
 $AC = (a^2 + a^2)^{1/2} = a(2)^{1/2}$



Therefore, $4r = a(2)^{1/2}$
 $\Rightarrow a = 4r / (2)^{1/2}$
 Thus, Volume of unit cell $= a^3 = \{ 4r / (2)^{1/2} \}^3$
 $a^3 = 64r^3 / 2(2)^{1/2} = 32r^3 / (2)^{1/2}$
 No. of unit cell in FCC = 4
 Volume of four spheres = $4 \times \pi r^3 (4/3)$
 Thus, packing efficiency = $[\pi r^3 (16/3)] / [32r^3 / (2)^{1/2}] = 0.74$ or 74 %

Q9. Silver crystallises in fcc lattice. If edge length of the cell is 4.07×10^{-8} cm and density is 10.5 g cm^{-3} , calculate the atomic mass of silver.

Solid State 12th Class Chemistry Solutions

Ramashish Paul



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Conceptual Chemistry Volume-I For Class XII S.K. Jain & Shailesh K. Jain, Conceptual Chemistry Volume I For Class XII

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Chemistry Vol.-1 YCT Expert Team , 2022-23 NTA NEET JEE MAIN Chemistry Vol 1 Chapter wise Solved Papers

Solid State Chemistry A. K. Cheetham, Peter Day, 1987

CBSE New Pattern Chemistry Class 12 for 2021-22 Exam (MCQs based book for Term 1) Abhishek Kaushik, Mahendra Rathod, 2021-09-10
1 This book deals with CBSE New Pattern Chemistry for Class 12. It is divided into 7 chapters as per Term 1 Syllabus. 3 Quick Revision Notes covering all the Topics of the chapter. 4 Carries all types of Multiple Choice Questions (MCQs). 5 Detailed Explanation for all types of questions. 6 3 practice papers based on entire Term 1 Syllabus with OMR Sheet. With the introduction of new exam pattern, CBSE has introduced 2 Term Examination Policy where Term 1 deals with MCQ based questions while Term 2 consists of Subjective Questions. Introducing Arihant's CBSE New Pattern Series, the first of its kind, providing the complete emphasis on Multiple Choice Questions which are designated in TERM 1 of each subject from Class 9th to 12th. Serving as a new preparatory guide, here we are presenting the all new edition of CBSE New Pattern Chemistry for Class 12 Term 1 that is designed to cover all the Term I chapters as per rationalized syllabus in a Complete Assertion Reasoning Based MCQs and Cased MCQs for the overall preparation. Detailed Explanations of the selected questions help students to get the pattern and questions as well. Lastly, 3 Practice Questions are provided for the revision of the concepts. TOC: Solid State, Solutions, p Block Elements, Haloalkanes and Haloarenes, Alcohols, Phenols, and Biomolecules. Practice Papers 1-3.

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