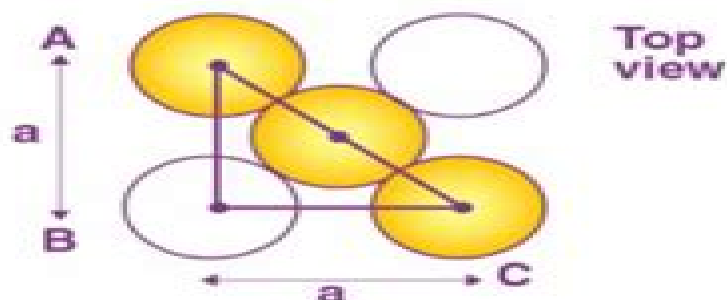


Let the edge length of the unit cell =  $a$   
 let the radius of each sphere =  $r$   
 Thus,  $AC = 4r$   
 From the right angled triangle  $ABC$ , we have:  
 $AC = (a^2 + a^2)^{1/2} = a(2)^{1/2}$



Therefore,  $4r = a(2)^{1/2}$   
 $\Rightarrow a = 4r / (2)^{1/2}$   
 Thus, Volume of unit cell  $= a^3 = \{ 4r / (2)^{1/2} \}^3$   
 $a^3 = 64r^3 / 2(2)^{1/2} = 32r^3 / (2)^{1/2}$   
 No. of unit cell in FCC = 4  
 Volume of four spheres =  $4 \times \pi r^3 (4/3)$   
 Thus, packing efficiency =  $[\pi r^3 (16/3)] / [32r^3 / (2)^{1/2}] = 0.74$  or 74 %

**Q9.** Silver crystallises in fcc lattice. If edge length of the cell is  $4.07 \times 10^{-8}$  cm and density is  $10.5 \text{ g cm}^{-3}$ , calculate the atomic mass of silver.

# **Solid State 12th Class Chemistry Solutions**

**James H. Crawford, Lawrence M. Slifkin**



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*CBSE Class 12th Chemistry (10 Most Likely Question Papers with Solution) By Career Point Kota* Career Point Kota,2020-07-28 The CBSE has made certain changes in the assessment structure from the session 2019 20 onwards In the new scheme of examination CCE and term system has been replaced with the Internal Assessment Single Annual Exam by CBSE itself Single exam conducted by CBSE will carry 80 marks whereas 20 marks are left to the schools for internal assessment CBSE has issued detailed guidelines on how the internal marks will be divided among different activities From 2019 onwards there will be internal choices in board examinations with increased internal options in the question paper Considering this change now a student has to prepare accordingly for board examinations The new assessment format brought with it excitement as well as anxiety And to help them prepare and excel in their CBSE board examination Career Point Kota has developed a series of 10 Most Likely Question Papers with Solutions The Key Features of Most Likely Question Papers with Solutions Series New OBJECTIVE TYPE question in each paper Syllabus of CBSE 2019 20 Based on the latest CBSE Syllabus Pattern Mind map of each chapter is given to visualize and help acquire a better understanding Important terms facts formulae and quick revision tips are given Covers questions asked in previous year board exams Toppers Answer Sheet as released by CBSE to understand the scoring technique We hope this book will gratify students need for the new CBSE pattern board exam and smoothen their path to success We wish to utilize the opportunity to place on record our special thanks to all the members of the Content Development team for their efforts to create this wonderful book

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Solid State Chemistry Elaine A. Moore, Lesley E. Smart, 2020-08-03 A comprehensive guide to solid state chemistry which is ideal for all undergraduate levels It covers well the fundamentals of the area from basic structures to methods of analysis but also introduces modern topics such as sustainability Dr Jennifer Readman University of Central Lancashire UK The latest edition of Solid State Chemistry combines clear explanations with a broad range of topics to provide students with a firm grounding in the major theoretical and practical aspects of the chemistry of solids Professor Robert Palgrave University College London UK Building a foundation with a thorough description of crystalline structures this fifth edition of Solid State Chemistry An Introduction presents a wide range of the synthetic and physical techniques used to prepare and characterise solids Going beyond this this largely nonmathematical introduction to solid state chemistry includes the bonding and electronic magnetic electrical and optical properties of solids Solids of particular interest porous solids superconductors and nanostructures are included Practical examples of applications and modern developments are given It offers students the opportunity to apply their knowledge in real life situations and will serve them well throughout their degree course New in the Fifth Edition A companion website which offers accessible resources for students and instructors alike featuring topics and tools such as quizzes videos web links and more A new chapter on sustainability in solid state chemistry written by an expert in this field Cryo electron microscopy X ray photoelectron spectroscopy ESCA Covalent organic frameworks Graphene oxide and bilayer graphene Elaine A Moore studied chemistry as an undergraduate at Oxford University and then stayed on to complete a DPhil in theoretical chemistry with Peter Atkins After a two year postdoctoral position at the University of Southampton she joined the Open University in 1975 becoming a lecturer in chemistry in 1977 senior lecturer in 1998 and reader in 2004 She retired in 2017 and currently has an honorary position at the Open University She has produced OU teaching texts in chemistry for courses at levels 1 2 and 3 and written texts in astronomy at level 2 and physics at level 3 She was team leader for the production and presentation of an Open University level 2 chemistry module delivered entirely online She is a Fellow of the Royal Society of Chemistry and a Senior Fellow of the Higher Education Academy She was co chair for the successful Departmental submission of an Athena Swan bronze award Lesley E Smart studied chemistry at Southampton University United Kingdom After completing a PhD in Raman spectroscopy she moved to a lectureship at the then Royal University of Malta After returning to the United Kingdom she took an SRC Fellowship to Bristol University to work on X ray crystallography From 1977 to 2009 she worked at the Open University chemistry department as a lecturer senior lecturer and Molecular Science Programme director and she held an honorary senior lectureship there until her death in 2016 At the Open University she was involved in the production of undergraduate courses in inorganic and physical chemistry and health sciences She served on the Council of the Royal Society of Chemistry and as the chair of their Benevolent Fund

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**Chemistry : Textbook For Class Xii** NCERT,2007-01-01

Treatise on Solid State Chemistry Norman Bruce Hannay,1976

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**Conceptual Chemistry Volume-I For Class XII** S.K. Jain & Shailesh K. Jain, Conceptual Chemistry Volume I For Class XII

*Chemistry Vol.-1* YCT Expert Team, 2022-23 NTA NEET JEE MAIN Chemistry Vol 1 Chapter wise Solved Papers

**CBSE New Pattern Chemistry Class 12 for 2021-22 Exam (MCQs based book for Term 1)** Abhishek Kaushik, Mahendra Rathod, 2021-09-10. This book deals with CBSE New Pattern Chemistry for Class 12. It is divided into 7 chapters as per Term 1 Syllabus. It includes 3 Quick Revision Notes covering all the topics of the chapter, 4 Carries all types of Multiple Choice Questions (MCQs), 5 Detailed Explanation for all types of questions, 6 3 practice papers based on entire Term 1 Syllabus with OMR Sheet. With the introduction of new exam pattern, CBSE has introduced 2 Term Examination Policy where Term 1 deals with MCQ based questions while Term 2 consists of Subjective Questions. Introducing Arihant's CBSE New Pattern Series, the first of its kind, providing complete emphasis on Multiple Choice Questions which are designated in TERM 1 of each subject from Class 9th to 12th. Serving as a new preparatory guide, here we are presenting the all new edition of CBSE New Pattern Chemistry for Class 12 Term 1 that is designed to cover all the Term I chapters as per rationalized syllabus in a Complete Assertion Reasoning Based MCQs and Cased MCQs for the overall preparation. Detailed Explanations of the selected questions help students to get the pattern and questions as well. Lastly, 3 Practice Questions are provided for the revision of the concepts. TOC: Solid State, Solutions, p Block Elements, Haloalkanes and Haloarenes, Alcohols, Phenols, and Biomolecules. Practice Papers 1-3.



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