

Intermolecular Forces Worksheet

Fill in the Blank:

Terms: bonds, charges, dipole, forces, dipole-dipole forces, hydrogen bonding, ions, ionic, London dispersion forces, unchanged, unequal

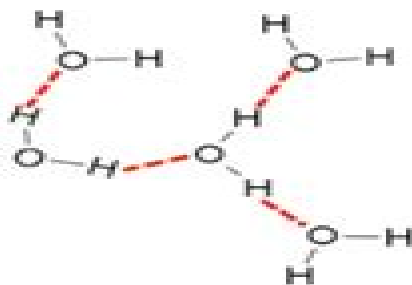
1. When a molecular substance changes state (ex: from liquid to gas), the atoms within the molecules are unchanged
2. No bonds are broken when a molecular substance changes state.
3. The forces holding ions together in ionic solids are ionic electrostatic forces. Opposite charges attract each other. These are the strongest intra-particle forces.
4. The strongest intermolecular forces in a sample of oxygen gas are the London Dispersion Forces
5. Unequal sharing of electrons in a molecule results in the formation of partial charges on the molecule.
6. A dipole occurs because one part of a molecule has a partial positive charge while another part of a molecule has a partial negative charge.

There are 3 intermolecular forces present in a sample of water. List the 3 intermolecular forces in order of strength from strongest to weakest:

1. Hydrogen Bonding
2. Dipole-Dipole Forces
3. London Dispersion Forces

Diagrams

Draw a diagram showing how 5 water molecules would interact in the liquid state:



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M Carnoy



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web later these 19 spatial features were reduced to two basic dimensions with the principal component analysis and the level of explanation of these dimensions on the overall school satisfaction was revealed by multiple regression analysis finding the level of satisfaction of parents with the spatial characteristics of primary schools differs

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web therefore spatial factors such as school buildings and gardens access to school and security issues should be a priority in terms of spatial planning since children in primary education need the custody and supervision of their parents the parents assessment of primary schools is of critical importance

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