

# Chemistry in 3D

Adding to What You Know

**Solubility Rules**

Sort through the material on page two and decide which compound is soluble and which compound is insoluble based on your solubility rules chart. If the compound is soluble place it in the beaker with the appropriate class. For example, if you get sodium chloride, you would put that into the beaker for chlorides since it is soluble. It would not go into the insoluble chlorides.

**Soluble**      **Insoluble**      **Soluble**      **Insoluble**

Acetates      Chlorides

Nitrates      Sulfates

Carbonates      Oxides

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## Solubility Rules Practice

By CoScience

For High School and Beyond

# 7h Solutions And Solubility Answers

**Effiong Eyo**

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# 7h Solutions And Solubility Answers

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