

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

A																	B	
												D	E					
C												H		K				
	F												I					
J	L	Z				M				N			O	P				
Q		R				S	T	U										
													V			W		
													X			Y		

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid

- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "Z" **C)** "K" **D)** A, B, and C **E)** A and B only

- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?

- _____ 4) Which letter labels the halogen element with the lowest atomic number?

- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"

- _____ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "M" **D)** "W" **E)** "Y"

continued on other side

Atoms And Periodic Table Review Answer Key

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Answer Key for The Mystery of the Periodic Table Staff of Christ the King Books,2018-02

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The Atom and the Periodic Table : Level 4 Chemistry Module SC 4.1 Algonquin College. Continuing Education. College and Career Preparation Department, Croll, Ann, 1995

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General Chemistry Key Concise Review Guide Sixth Edition Darrell D. Ebbing,1999

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Table of Contents Atoms And Periodic Table Review Answer Key

1. Understanding the eBook Atoms And Periodic Table Review Answer Key
 - The Rise of Digital Reading Atoms And Periodic Table Review Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Atoms And Periodic Table Review Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Atoms And Periodic Table Review Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Atoms And Periodic Table Review Answer Key
 - Personalized Recommendations
 - Atoms And Periodic Table Review Answer Key User Reviews and Ratings
 - Atoms And Periodic Table Review Answer Key and Bestseller Lists
5. Accessing Atoms And Periodic Table Review Answer Key Free and Paid eBooks
 - Atoms And Periodic Table Review Answer Key Public Domain eBooks
 - Atoms And Periodic Table Review Answer Key eBook Subscription Services
 - Atoms And Periodic Table Review Answer Key Budget-Friendly Options
6. Navigating Atoms And Periodic Table Review Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Atoms And Periodic Table Review Answer Key Compatibility with Devices
 - Atoms And Periodic Table Review Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Atoms And Periodic Table Review Answer Key
 - Highlighting and Note-Taking Atoms And Periodic Table Review Answer Key
 - Interactive Elements Atoms And Periodic Table Review Answer Key
8. Staying Engaged with Atoms And Periodic Table Review Answer Key

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Atoms And Periodic Table Review Answer Key
9. Balancing eBooks and Physical Books Atoms And Periodic Table Review Answer Key
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Atoms And Periodic Table Review Answer Key
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Atoms And Periodic Table Review Answer Key
- Setting Reading Goals Atoms And Periodic Table Review Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Atoms And Periodic Table Review Answer Key
- Fact-Checking eBook Content of Atoms And Periodic Table Review Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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