

CHEMISTRY 1B MIDTERM STUDY GUIDE

Name: _____

STOICHIOMETRY:

1. What is a limiting reactant? _____
2. What is an excess reactant? _____
3. Zinc and sulfur react to form zinc sulfide according to the equation.



If 25.0 g of zinc and 30.0 g of sulfur are mixed,

- a. Which chemical is the limiting reactant?
- b. How many grams of ZnS will be formed?
- c. How many grams of the excess reactant will remain after the reaction is over?

GAS LAWS:

1. The 4 measurable properties of gases. _____
2. The units for volume are _____. The units for temperature are _____. The units for pressure are _____.
3. What does STP mean? What is the value of temperature? What is the value of pressure?
4. Know the relationship between pressure and volume: As the pressure increases, volume _____.
- a. A weather balloon with a volume of 1.375 L is released from the Earth's surface at sea level(100kPa.) What volume will the balloon occupy at an altitude where the pressure is 10kPa (1kPa = 1000Pa)? **14 L**
5. Know the relationship between temperature and volume. As the temperature increases, volume _____.
- a. A balloon filled with oxygen gas occupies a volume of 5.5 L at 25C. What will the volume be at 100C (1000mL = 1L)? **6.9L**
6. Know the relationship between temperature and pressure (direct or indirect). As the temperature increases, temperature _____.

Chemistry Midterm Study Guide

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