

Isotope Notation

Chem Worksheet 4-2

Name Solutions

Any given element can have more than one isotope. To distinguish between the different isotopes of an atom, the element is named with its mass number, for example lithium-7. Remember that the **mass number** is the number of protons and neutrons. When symbols are used to represent an isotope the mass number is written next to the symbol on the top left. The atomic number is written on the bottom left. Recall that the **atomic number** is the number of protons.

EXAMPLES



Answer the following questions about atoms.

- The identity of an atom is determined by the number of protons.
- The particle(s) found inside the nucleus are called: protons - neutrons.
- The number of protons and neutrons combined is called the mass number.
- In large atoms the number of protons is less than the number of neutrons.
- The number of protons is also called the atomic number.
- Isotopes have the same number of protons, but different numbers of neutrons.
- The number of protons found in a sulfur atom is 16.
- The number of neutrons found in an aluminum-27 atom is 14.
- The number of electrons found in a zinc atom is 30.
- What is the name of the element with 82 protons? lead.

Give the symbols for the nuclides described by the following particles. Include the atomic number and the mass number.

- | | |
|---|---|
| 11. 92 protons, 145 neutrons: ²³⁷ ₉₂ U | 15. 20 protons, 20 neutrons: ⁴⁰ ₂₀ Ca |
| 12. 8 protons, 10 neutrons: ¹⁸ ₈ O | 16. 22 protons, 23 neutrons: ⁴⁵ ₂₂ Ti |
| 13. 82 protons, 125 neutrons: ²⁰⁷ ₈₂ Pb | 17. 18 protons, 22 neutrons: ⁴⁰ ₁₈ Ar |
| 14. 80 protons, 119 neutrons: ¹⁹⁹ ₈₀ Hg | 18. 25 protons, 32 neutrons: ⁵⁷ ₂₅ Mn |

Determine the number of protons and neutrons from the following symbols.

- | | | |
|---|--|--|
| 19. ¹⁰ ₃ B p ⁺ =5 n ⁰ =5 | 23. ¹⁶² ₆₆ Dy p ⁺ =66 n ⁰ =99 | 27. ¹²⁸ ₅₂ Te p ⁺ =52 n ⁰ =74 |
| 20. ¹³ ₇ N p ⁺ =7 n ⁰ =8 | 24. ⁵⁶ ₂₆ Fe p ⁺ =26 n ⁰ =30 | 28. ³⁵ ₁₇ Cl p ⁺ =17 n ⁰ =18 |
| 21. ⁷⁸ ₃₄ Se p ⁺ =34 n ⁰ =45 | 25. ¹⁵¹ ₆₂ Sm p ⁺ =62 n ⁰ =89 | 29. ¹⁰⁸ ₄₇ Ag p ⁺ =47 n ⁰ =60 |
| 22. ¹¹⁸ ₅₀ Sn p ⁺ =50 n ⁰ =69 | 26. ¹⁹⁴ ₇₈ Pt p ⁺ =78 n ⁰ =117 | 30. ⁹³ ₄₁ Niobium-93 p ⁺ =41 n ⁰ =52 |

Atomic Structure Teachers Answers

Joseph Seibert Butterweck



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with elementary students through a research based foundation *Learning Elementary Science Class 8 Teacher Resource Book (Academic Year 2023-24)* ,2023-05-20 *Learning Elementary Science Class 8 Teacher Resource Book Academic Year 2023 24* *Learning Elementary Physics Class 8 Teacher Resource Book (Academic Year 2023-24)* ,2023-05-20 *Learning Elementary Physics Class 8 Teacher Resource Book Academic Year 2023 24* *Software for Teaching Science* Roger Frost,1998 Teaching and Learning STEM Richard M. Felder,Rebecca Brent,2016-02-22 Rethink traditional teaching methods to improve student learning and retention in STEM Educational research has repeatedly shown that compared to traditional teacher centered instruction certain learner centered methods lead to improved learning outcomes greater development of critical high level skills and increased retention in science technology engineering and mathematics STEM disciplines *Teaching and Learning STEM* presents a trove of practical research based strategies for designing and teaching STEM courses at the university community college and high school levels The book draws on the authors extensive backgrounds and decades of experience in STEM education and faculty development Its engaging and well illustrated descriptions will equip you to implement the strategies in your courses and to deal effectively with problems including student resistance that might occur in the implementation The book will help you Plan and conduct class sessions in which students are actively engaged no matter how large the class is Make good use of technology in face to face online and hybrid courses and flipped classrooms Assess how well students are acquiring the knowledge skills and conceptual understanding the course is designed to teach Help students develop expert problem solving skills and skills in communication creative thinking critical thinking high performance teamwork and self directed learning Meet the learning needs of STEM students with a broad diversity of attributes and backgrounds The strategies presented in *Teaching and Learning STEM* don t require revolutionary time intensive changes in your teaching but rather a gradual integration of traditional and new methods The result will be continual improvement in your teaching and your students learning More information about *Teaching and Learning STEM* can be found at <http://educationdesignsinc.com> book including its preface foreword table of contents first chapter a reading guide and reviews in 10 prominent STEM education journals **OCR A Chemistry AS Student Unit Guide: F321 New Edition Atoms, Bonds and Groups** Mike Smith,2012-08-31 Written by a former senior examiner Mike Smith this OCR A AS Chemistry Student Unit Guide is the essential study companion for Unit F321 Atoms Bonds and Groups This full colour book includes all you need to know to prepare for your unit exam clear guidance on the content of the unit with topic summaries knowledge check questions and a quick reference index examiner s advice throughout so you will know what to expect in the exam and will be able to demonstrate the skills required exam style questions with graded student responses so you can see clearly what is required to get a better grade **Resources in Education** ,1998 The Problem of Teaching High School Pupils how to Study Joseph Seibert Butterweck,1926 *Voyage of Adventure. Annotated Teacher's Edition* ,1986

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