

Name: Vivek Shanavas  
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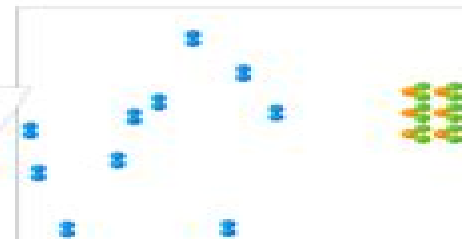
**Vocabulary:** activated complex, catalyst, chemical reaction, concentration, enzyme, half-life, molecule, product, reactant, surface area

**Prior Knowledge Questions** (Do these BEFORE using the Gizmo.)

- Suppose you added a spoonful of sugar to hot water and another to ice-cold water. Which type of water will cause the sugar to dissolve more quickly? The hot water
- Suppose you held a lighted match to a solid hunk of wood and another match to a pile of wood shavings. Which form of wood will catch fire more easily? The shavings

**Gizmo Warm-up**

A **chemical reaction** causes the chemical compositions of substances to change. **Reactants** are substances that enter into a reaction. **Products** are substances produced in the reaction. **Theory** Gizmo does not experiment with factors that influence the rate at which reactants are transformed into products in a chemical reaction.



You will need blue, green, and orange markers or colored pencils for the first part of this activity.

- Look at the key at the bottom of the SIMULATION pane. In the space below, draw the two reactants and two products of this chemical reaction.



- Click Play (▶) and observe what you see?

*Reactant A is a diatomic molecule and Reactant B is a diatomic molecule. When Reactant A succeeds in breaking the bonds of Reactant B, the orange substance on Reactant B breaks off and bonds with Reactant A forming Product B, leaving the two green substance molecules which are Product A.*



# Collision Theory Lab Gizmo Answer Key

**Lei Shi**



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