

Chapter 1: Periodic Trends (NB p. 29-30)

Vocabulary: Period, group, atomic radius, ionization energy.

Equations & Constants: N/A

Practice Problems:

D 1. The energy required to remove an electron from an atom is the atom's
a. electron affinity.
b. electron energy.
c. electronegativity.
d. ionization energy.

D 2. The element that has the greatest electronegativity is
a. oxygen.
b. sodium.
c. chlorine.
d. fluorine.

A 3. One-half the distance between the nuclei of identical atoms that are bonded together is called the
a. atomic radius.
b. atomic diameter.
c. atomic volume.
d. electron cloud.

C 4. A positive ion is known as a(n)
a. ionic radius.
b. valence electron.
c. cation.
d. anion.

D 5. A negative ion is known as a(n)
a. ionic radius.
b. valence electron.
c. cation.
d. anion.

A 6. Across a period in the periodic table, atomic radii
a. gradually decrease.
b. gradually decrease, then sharply increase.
c. gradually increase.
d. gradually increase, then sharply decrease.

C 7. Which of the following elements has the largest atomic radius?
a. Fluorine
b. Bromine
c. Iodine
d. Chlorine

A 8. Which of the following elements has the largest atomic radius?
a. Chromium
b. Manganese
c. Zinc
d. Iron

C 9. Which of the following elements has the largest atomic radius?
a. Calcium
b. Magnesium
c. Barium
d. Strontium

D 10. Which of the following elements has the largest ionization energy?
a. Hydrogen
b. Helium
c. Lithium
d. Beryllium

A 11. Which of the following elements has the largest ionization energy?
a. Phosphorous
b. Arsenic
c. Selenium
d. Carbon

B 12. Which of the following elements has the largest ionization energy?
a. Calcium
b. Magnesium
c. Barium
d. Strontium

Atomic Radius

1. Atoms get **larger** going **down a group** because **adding new shells/energy levels**

2. Atoms get **smaller** going **across a period** because **not adding energy levels but \uparrow Z_{eff}**

Continental Academy Test Answers

William W. Brickman



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<i>The Encyclopædia Britannica: Evangelical Church-Francis Joseph</i> ,1910
<i>Encyclopedia Britannica</i> Hugh Chisholm,1911
Vegetation Description and Data Analysis Martin Kent,2011-11-14 Vegetation Description and Data Analysis A Practical Approach Second Edition is a fully revised and up dated edition of this key text The book takes account of recent advances in the field whilst retaining the original reader friendly approach to the coverage of vegetation description and multivariate analysis in the context of vegetation data and plant ecology Since the publication of the hugely popular first edition there have been significant developments in computer hardware and software new key journals have been established in the field and scope and application of vegetation description and analysis has become a truly global field This new edition includes full coverage of new developments and technologies This contemporary and comprehensive edition of this well known and respected textbook will prove invaluable to undergraduate and graduate students in biological sciences environmental science geography botany agriculture forestry and biological conservation Fully international approach Includes illustrative case studies throughout Now with new material on the nature of plant communities transitional areas between plant communities induction and deduction of plant ecology diversity indices and dominance diversity curves multivariate analysis in ecology Accessible reader friendly style Now with new and improved illustrations
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Table of Contents Continental Academy Test Answers

1. Understanding the eBook Continental Academy Test Answers
 - The Rise of Digital Reading Continental Academy Test Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Continental Academy Test Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Continental Academy Test Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Continental Academy Test Answers
 - Personalized Recommendations
 - Continental Academy Test Answers User Reviews and Ratings
 - Continental Academy Test Answers and Bestseller Lists

5. Accessing Continental Academy Test Answers Free and Paid eBooks
 - Continental Academy Test Answers Public Domain eBooks
 - Continental Academy Test Answers eBook Subscription Services
 - Continental Academy Test Answers Budget-Friendly Options
6. Navigating Continental Academy Test Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Continental Academy Test Answers Compatibility with Devices
 - Continental Academy Test Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Continental Academy Test Answers
 - Highlighting and Note-Taking Continental Academy Test Answers
 - Interactive Elements Continental Academy Test Answers
8. Staying Engaged with Continental Academy Test Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Continental Academy Test Answers
9. Balancing eBooks and Physical Books Continental Academy Test Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Continental Academy Test Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Continental Academy Test Answers
 - Setting Reading Goals Continental Academy Test Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Continental Academy Test Answers
 - Fact-Checking eBook Content of Continental Academy Test Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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