

Dilutions Worksheet - Solutions

- 1) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?

$$\begin{aligned}M_1 V_1 &= M_2 V_2 \\(0.15 M)(125 \text{ mL}) &= x (150 \text{ mL}) \\x &= 0.125 M\end{aligned}$$

- 2) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?

$$\begin{aligned}M_1 V_1 &= M_2 V_2 \\(0.15 M)(100 \text{ mL}) &= x (150 \text{ mL}) \\x &= 0.100 M\end{aligned}$$

- 3) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

$$\begin{aligned}M_1 V_1 &= M_2 V_2 \\(10 M)(250 \text{ mL}) &= (0.05 M) x \\x &= 50,000 \text{ mL}\end{aligned}$$

- 4) I have 345 mL of a 1.5 M NaCl solution. If I boil the water until the volume of the solution is 250 mL, what will the molarity of the solution be?

$$\begin{aligned}M_1 V_1 &= M_2 V_2 \\(1.5 M)(345 \text{ mL}) &= x (250 \text{ mL}) \\x &= 2.07 M\end{aligned}$$

- 5) How much water would I need to add to 500 mL of a 2.4 M KCl solution to make a 1.0 M solution?

$$\begin{aligned}M_1 V_1 &= M_2 V_2 \\(2.4 M)(500 \text{ mL}) &= (1.0 M) x \\x &= 1200 \text{ mL}\end{aligned}$$

1200 mL will be the final volume of the solution. However, since there's already 500 mL of solution present, you only need to add 700 mL of water to get 1200 mL as your final volume. The answer: 700 mL.

Solutions Worksheet 2 Molarity And Dilution Problems

Rosina Ehmman



Solutions Worksheet 2 Molarity And Dilution Problems

Linne & Ringsrud's Clinical Laboratory Science - E-Book Mary Louise Turgeon,2015-02-10 Using a discipline by discipline approach Linne perforated pages make it easy for students to remove for use in the lab and for assignment of review questions as homework NEW Instrumentation updates show new technology being used in the lab NEW Additional key terms in each chapter cover need to know terminology NEW Additional tables and figures in each chapter clarify clinical lab science concepts *Chemistry 2e* Paul Flowers,Richard Langely,William R. Robinson,Klaus Hellmut

Theopold,2019-02-14 *Chemistry 2e* is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in *Chemistry 2e* are described in the preface to help instructors transition to the second edition *Holt Chemistry*,2003 *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998

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Problems of Instrumental Analytical Chemistry (Second Edition) Jose Manuel Andrade-Garda, 2023-10-30
The book is intended as a tool for undergraduate students As European Universities converged to the EEES Bologna space the time available for theoretical and practical lessons at the classroom has reduced dramatically The fundamental principles and basic theory of Analytical Chemistry are covered by many wonderful textbooks but the numerical part is not so Although it is true that many of them present some worked examples the students need more support Indeed many teachers observed a serious decline in the calculation capabilities of their students This book is intended to help undergraduate students of Instrumental Analytical Chemistry to develop strategies to generate information from experimental results in an efficient and reliable way The exercises will provide standard protocols that students can follow to address the most common calculation steps required in laboratory daily work It is assumed that they will use the hands on guide after the basic principles of the analytical techniques were presented in their classes Easy to follow diagrams are included to facilitate understanding of the calculations and avoid common errors As a novelty QR codes are inserted into the text to offer additional extra information and or links to reputed websites with additional explanations and or computer animations This new feature is a distinctive one very rare in scientific or teaching oriented books

Calculations for Molecular Biology and Biotechnology Frank H. Stephenson, 2010-07-30
Calculations for Molecular Biology and Biotechnology A Guide to Mathematics in the Laboratory Second Edition provides an introduction to the myriad of laboratory calculations used in molecular biology and biotechnology The book begins by

discussing the use of scientific notation and metric prefixes which require the use of exponents and an understanding of significant digits It explains the mathematics involved in making solutions the characteristics of cell growth the multiplicity of infection and the quantification of nucleic acids It includes chapters that deal with the mathematics involved in the use of radioisotopes in nucleic acid research the synthesis of oligonucleotides the polymerase chain reaction PCR method and the development of recombinant DNA technology Protein quantification and the assessment of protein activity are also discussed along with the centrifugation method and applications of PCR in forensics and paternity testing Topics range from basic scientific notations to complex subjects like nucleic acid chemistry and recombinant DNA technology Each chapter includes a brief explanation of the concept and covers necessary definitions theory and rationale for each type of calculation Recent applications of the procedures and computations in clinical academic industrial and basic research laboratories are cited throughout the text New to this Edition Updated and increased coverage of real time PCR and the mathematics used to measure gene expression More sample problems in every chapter for readers to practice concepts

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Solutions Worksheet 2 Molarity And Dilution Problems Introduction

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