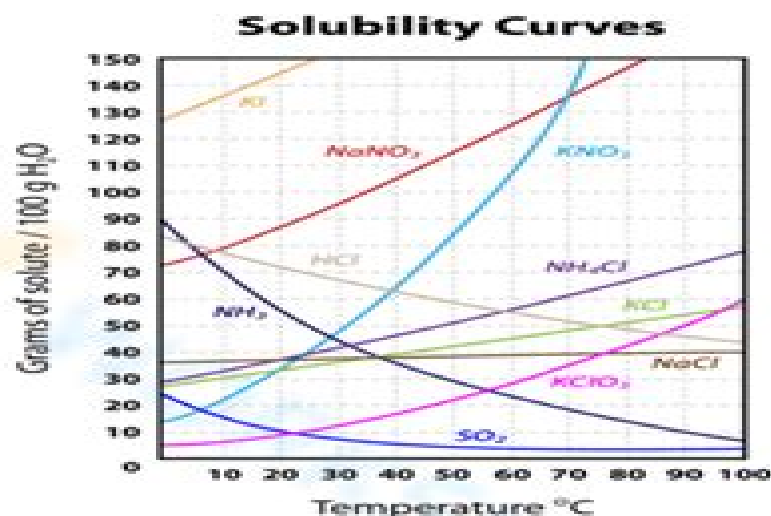


WS 6.1: Interpreting Solubility Curves

Directions: Use the Solubility Curve Table below to answer all questions. So you do not have to keep flipping your paper over you might want to put the solubility curve on your phone or computer screen when answering the questions on the back. All questions assume the amount of water is 100 ml.



- What is the solvent in this Solubility Curve?
water
- Saturation points are given per how much solvent?
100 ml or 100 g
- What is the saturation point of NaNO₃ at 10°C, 40°C, & 80°C?
@ 10°C = 75 g
@ 40°C = 105 g
@ 80°C = 145 g
- How much NH₃ can you dissolve in water at 10°C, 30°C, & 90°C?
@ 10°C = 70 g
@ 30°C = 55 g
@ 90°C = 5 g
- What is the maximum amount of KCl that can be dissolved in water at 20°C, 50°C, & 90°C?
@ 20°C = 35 g
@ 50°C = 35 g
@ 90°C = 35 g
- If you want to make a saturated solution of NaCl how many grams would you dissolve in 100 grams of water @ 80°C?
About 40 grams
- Except for KI What substance is the most soluble in water at 20°C?
NaNO₃
- Except for KI, what substance is the most soluble in water at 72°C?
KNO₃
- Look up The boiling points for SO₂, HCl and NH₃ and write them below:
B.P. SO₂ = -10°C
B.P. HCl = -85°C
B.P. NH₃ = -33°C
- Based on the boiling points you looked up, are SO₂, HCl & NH₃ solids, liquids, or gases at 0°C? (and anywhere else on the solubility curve)
gases

Interpreting Solubility Curves Packet Answers

Elissa Huddart



Interpreting Solubility Curves Packet Answers

Solubility Curves William Henry Mickley,1933 Solubility curves Charles Hertz Frantz,1933 **Resources for Teaching Middle School Science** Smithsonian Institution,National Academy of Engineering,National Science Resources Center of the National Academy of Sciences,Institute of Medicine,1998-04-30 With age appropriate inquiry centered curriculum materials and sound teaching practices middle school science can capture the interest and energy of adolescent students and expand their understanding of the world around them Resources for Teaching Middle School Science developed by the National Science Resources Center NSRC is a valuable tool for identifying and selecting effective science curriculum materials that will engage students in grades 6 through 8 The volume describes more than 400 curriculum titles that are aligned with the National Science Education Standards This completely new guide follows on the success of Resources for Teaching Elementary School Science the first in the NSRC series of annotated guides to hands on inquiry centered curriculum materials and other resources for science teachers The curriculum materials in the new guide are grouped in five chapters by scientific area Physical Science Life Science Environmental Science Earth and Space Science and Multidisciplinary and Applied Science They are also grouped by type core materials supplementary units and science activity books Each annotation of curriculum material includes a recommended grade level a description of the activities involved and of what students can be expected to learn a list of accompanying materials a reading level and ordering information The curriculum materials included in this book were selected by panels of teachers and scientists using evaluation criteria developed for the guide The criteria reflect and incorporate goals and principles of the National Science Education Standards The annotations designate the specific content standards on which these curriculum pieces focus In addition to the curriculum chapters the guide contains six chapters of diverse resources that are directly relevant to middle school science Among these is a chapter on educational software and multimedia programs chapters on books about science and teaching directories and guides to science trade books and periodicals for teachers and students Another section features institutional resources One chapter lists about 600 science centers museums and zoos where teachers can take middle school students for interactive science experiences Another chapter describes nearly 140 professional associations and U S government agencies that offer resources and assistance Authoritative extensive and thoroughly indexed and the only guide of its kind Resources for Teaching Middle School Science will be the most used book on the shelf for science teachers school administrators teacher trainers science curriculum specialists advocates of hands on science teaching and concerned parents

Chemistry 2e Paul Flowers,Richard Langely,William R. Robinson,Klaus Hellmut Theopold,2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to

their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition

Strengthening Forensic Science in the United States National Research Council, Division on Engineering and Physical Sciences, Committee on Applied and Theoretical Statistics, Policy and Global Affairs, Committee on Science, Technology, and Law, Committee on Identifying the Needs of the Forensic Sciences Community, 2009-07-29 Scores of talented and dedicated people serve the forensic science community performing vitally important work However they are often constrained by lack of adequate resources sound policies and national support It is clear that change and advancements both systematic and scientific are needed in a number of forensic science disciplines to ensure the reliability of work establish enforceable standards and promote best practices with consistent application Strengthening Forensic Science in the United States A Path Forward provides a detailed plan for addressing these needs and suggests the creation of a new government entity the National Institute of Forensic Science to establish and enforce standards within the forensic science community The benefits of improving and regulating the forensic science disciplines are clear assisting law enforcement officials enhancing homeland security and reducing the risk of wrongful conviction and exoneration Strengthening Forensic Science in the United States gives a full account of what is needed to advance the forensic science disciplines including upgrading of systems and organizational structures better training widespread adoption of uniform and enforceable best practices and mandatory certification and accreditation programs While this book provides an essential call to action for congress and policy makers it also serves as a vital tool for law enforcement agencies criminal prosecutors and attorneys and forensic science educators

Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards

Fitting Models to Biological Data Using Linear and Nonlinear Regression Harvey Motulsky, Arthur Christopoulos, 2004-05-27 Most biologists use nonlinear regression more than any other statistical technique but there are very few places to learn about curve fitting This book by the author of the very successful Intuitive

Biostatistics addresses this relatively focused need of an extraordinarily broad range of scientists **Chemistry**, 2015-03-16

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Quantitative Chemical Analysis Daniel C. Harris, Chuck Lucy, 2015-05-29 The gold standard in analytical chemistry Dan Harris Quantitative Chemical Analysis provides a sound physical understanding of the principles of analytical chemistry and their applications in the disciplines Pearson Chemistry Queensland 11 Skills and Assessment Book Elissa Huddart, 2018-10-04 Introducing the Pearson Chemistry 11 Queensland Skills and Assessment Book Fully aligned to the new QCE 2019 Syllabus Write in Skills and Assessment Book written to support teaching and learning across all requirements of the new Syllabus providing practice application and consolidation of learning Opportunities to apply and practice performing calculations and using algorithms are integrated throughout worksheets practical activities and question sets All activities are mapped from the Student Book at the recommend point of engagement in the teaching program making integration of practice and rich learning activities a seamless inclusion Developed by highly experienced and expert author teams with lead

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Applied Engineering Principles Manual - Training Manual (NAVSEA) Naval Sea Systems Command,2019-07-15 Chapter 1 ELECTRICAL REVIEW 1 1 Fundamentals Of Electricity 1 2 Alternating Current Theory 1 3 Three Phase Systems And Transformers 1 4 Generators 1 5 Motors 1 6 Motor Controllers 1 7 Electrical Safety 1 8 Storage Batteries 1 9 Electrical Measuring Instruments Chapter 2 ELECTRONICS REVIEW 2 1 Solid State Devices 2 2 Magnetic Amplifiers 2 3 Thermocouples 2 4 Resistance Thermometry 2 5 Nuclear Radiation Detectors 2 6 Nuclear Instrumentation Circuits 2 7 Differential Transformers 2 8 D C Power Supplies 2 9 Digital Integrated Circuit Devices 2 10 Microprocessor Based Computer Systems Chapter 3 REACTOR THEORY REVIEW 3 1 Basics 3 2 Stability Of The Nucleus 3 3 Reactions 3 4 Fission 3 5 Nuclear Reaction Cross Sections 3 6 Neutron Slowing Down 3 7 Thermal Equilibrium 3 8 Neutron Density Flux Reaction Rates And Power 3 9 Slowing Down Diffusion And Migration Lengths 3 10 Neutron Life Cycle And The Six Factor Formula 3 11 Buckling Leakage And Flux Shapes 3 12 Multiplication Factor 3 13 Temperature Coefficient *Basic Concepts in*

Biochemistry: A Student's Survival Guide Hiram F. Gilbert,2000 Basic Concepts in Biochemistry has just one goal to review the toughest concepts in biochemistry in an accessible format so your understanding is through and complete BOOK JACKET

CK-12 Chemistry - Second Edition CK-12 Foundation,2011-10-14 CK 12 Foundation s Chemistry Second Edition FlexBook covers the following chapters Introduction to Chemistry scientific method history Measurement in Chemistry measurements formulas Matter and Energy matter energy The Atomic Theory atom models atomic structure sub atomic particles The Bohr Model of the Atom electromagnetic radiation atomic spectra The Quantum Mechanical Model of the Atom energy standing waves Heisenberg Schrodinger The Electron Configuration of Atoms Aufbau principle electron configurations Electron Configuration and the Periodic Table electron configuration position on periodic table Chemical Periodicity atomic size ionization energy electron affinity Ionic Bonds and Formulas ionization ionic bonding ionic compounds Covalent Bonds and Formulas nomenclature electronic molecular geometries octet rule polar molecules The Mole Concept formula stoichiometry Chemical Reactions balancing equations reaction types Stoichiometry limiting reactant equations yields heat of reaction The Behavior of Gases molecular structure properties combined gas law universal gas law Condensed Phases Solids and Liquids intermolecular forces of attraction phase change phase diagrams Solutions and Their Behavior concentration solubility colligate properties dissociation ions in solution Chemical Kinetics reaction rates factors that affect rates Chemical Equilibrium forward reverse reaction rates equilibrium constant Le Chatelier s principle solubility product constant Acids Bases strong weak acids and bases hydrolysis of salts pHNeutralization dissociation of water acid base indicators acid base titration buffers Thermochemistry bond breaking formation heat of reaction formation Hess law entropy Gibb s free energy Electrochemistry oxidation reduction electrochemical cells Nuclear Chemistry radioactivity nuclear equations nuclear energy Organic Chemistry straight chain aromatic hydrocarbons functional groups Chemistry Glossary **Polymer Solutions** Iwao

Teraoka, 2004-04-07 *Polymer Solutions An Introduction to Physical Properties* offers a fresh inclusive approach to teaching the fundamentals of physical polymer science. Students, instructors, and professionals in polymer chemistry, analytical chemistry, organic chemistry, engineering, materials, and textiles will find Iwao Teraoka's text at once accessible and highly detailed in its treatment of the properties of polymers in the solution phase. Teraoka's purpose in writing *Polymer Solutions* is twofold: to familiarize the advanced undergraduate and beginning graduate student with basic concepts, theories, models, and experimental techniques for polymer solutions and to provide a reference for researchers working in the area of polymer solutions as well as those in charge of chromatographic characterization of polymers. The author's incorporation of recent advances in the instrumentation of size exclusion chromatography, the method by which polymers are analyzed, renders the text particularly topical. Subjects discussed include Real, ideal, Gaussian, semirigid, and branched polymer chains; Polymer solutions and thermodynamics; Static light scattering of a polymer solution; Dynamic light scattering and diffusion of polymers; Dynamics of dilute and semidilute polymer solutions. Study questions at the end of each chapter not only provide students with the opportunity to test their understanding but also introduce topics relevant to polymer solutions not included in the main text. With over 250 geometrical model diagrams, *Polymer Solutions* is a necessary reference for students and for scientists pursuing a broader understanding of polymers.

How Tobacco Smoke Causes Disease United States. Public Health Service. Office of the Surgeon General, 2010. This report considers the biological and behavioral mechanisms that may underlie the pathogenicity of tobacco smoke. Many Surgeon General's reports have considered research findings on mechanisms in assessing the biological plausibility of associations observed in epidemiologic studies. Mechanisms of disease are important because they may provide plausibility, which is one of the guideline criteria for assessing evidence on causation. This report specifically reviews the evidence on the potential mechanisms by which smoking causes diseases and considers whether a mechanism is likely to be operative in the production of human disease by tobacco smoke. This evidence is relevant to understanding how smoking causes disease, to identifying those who may be particularly susceptible, and to assessing the potential risks of tobacco products.

White and Pharoah's Oral Radiology Sanjay Mallya, Ernest Lam, 2018-09-12. Written specifically for dentists, *White and Pharoah's Oral Radiology: Principles and Interpretation*, 8th Edition, incorporates over 1,500 high-quality radiographic images and illustrations to demonstrate core concepts and essential principles and techniques of oral and maxillofacial radiology. The new edition of this bestselling book delivers with state-of-the-art information on oral radiology principles and techniques and image interpretation. Dental students will gain a solid foundation in radiation physics, radiation biology, and radiation safety and protection before introducing specialized techniques such as MRI and CT. As well, students will learn how to recognize the key radiographic features of pathologic conditions and interpret radiographs accurately. The 8th edition also includes new chapters on Radiologic Anatomy Beyond 3D Imaging and Diseases Affecting the Structure of Bone. A practical guide to using today's technology, this unique text helps your students provide state-of-the-art

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