

Model 2 – Ternary Acids (Oxyacids)

Acid	Name of Acid in Aqueous Solution	Cation (+)	Polyatomic Anion (-)	Polyatomic Anion Name
HClO_3	Chloric acid	H_3O^+	ClO_3^-	Chlorate
H_2SO_3	Sulfurous acid	$2\text{H}_3\text{O}^+$	SO_3^-	Sulfite
H_2SO_4	Sulfuric acid	$2\text{H}_3\text{O}^+$	SO_4^-	Sulfate
H_3PO_3	Phosphorous acid	$3\text{H}_3\text{O}^+$	PO_3^-	Phosphite
H_3PO_4	Phosphoric acid	$3\text{H}_3\text{O}^+$	PO_4^-	Phosphate
HNO_3	Nitric acid	H_3O^+	NO_3^-	Nitrate
HNO_2	Nitrous acid	H_3O^+	NO_2^-	Nitrite
H_2CO_3	Carbonic acid	$2\text{H}_3\text{O}^+$	CO_3^-	Carbonate

6. Look at the formulas of the ternary acids in Model 2.

a. How are ternary acids different from binary acids in their structure?

Ternary acids contain three elements and they also contain the element oxygen.

b. What number do you think the prefix "ter-" refers to?

"ter" refers to three

7. When ternary acids are mixed with water, ions will form. Fill in the table above with the formulas and names of the anions.

8. Examine the pairs of ternary acids in Model 2 that contain sulfur, phosphorus, and nitrogen. Each pair has one acid that ends in "-ic" and another that ends in "-ous." These endings are related to the name of the polyatomic anion found in the acid ("-ate" or "-ite"). Complete the statements below with the correct acid name ending.

Polyatomic anion ending is "-ate" → acid name ending is "ic"

Polyatomic anion ending is "-ite" → acid name ending is "ous"

9. If the prefix "hydro-" were used to name a ternary acid, what problem would this create when naming HClO_3 ?

ClO_3^- has the name of chlorate. "ate" must be replaced with "ic". If "hydro" is used as a prefix, it would be hydrochloric acid. This has the formula HCl , not HClO_3 .

10. Write a rule for naming ternary acids.

*- The prefix "hydro" is not used in the name when an acid contains more than two elements and one of the elements is oxygen.
- "ate" is replaced with "ic" and "ite" is replaced with "ous".*

11. Predict the formula for chlorous acid.



12. Circle the acid(s) below that would be named beginning with the prefix "hydro-."



Naming Acids Pogil Packet Answers

**Neil A. Campbell, Jane B. Reece, Fred
W. Holtzclaw, Theresa Knapp Holtzclaw**

Naming Acids Pogil Packet Answers

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POGIL Activities for High School Chemistry High School POGIL Initiative, 2012

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Pulmonary Gas Exchange G. Kim Prisk,Susan R. Hopkins,2013-08-01 The lung receives the entire cardiac output from the right heart and must load oxygen onto and unload carbon dioxide from perfusing blood in the correct amounts to meet the metabolic needs of the body It does so through the process of passive diffusion Effective diffusion is

accomplished by intricate parallel structures of airways and blood vessels designed to bring ventilation and perfusion together in an appropriate ratio in the same place and at the same time Gas exchange is determined by the ventilation perfusion ratio in each of the gas exchange units of the lung In the normal lung ventilation and perfusion are well matched and the ventilation perfusion ratio is remarkably uniform among lung units such that the partial pressure of oxygen in the blood leaving the pulmonary capillaries is less than 10 Torr lower than that in the alveolar space In disease the disruption to ventilation perfusion matching and to diffusional transport may result in inefficient gas exchange and arterial hypoxemia This volume covers the basics of pulmonary gas exchange providing a central understanding of the processes involved the interactions between the components upon which gas exchange depends and basic equations of the process The Double Helix James D. Watson, 1969-02 Since its publication in 1968 The Double Helix has given countless readers a rare and exciting look at one highly significant piece of scientific research Watson and Crick's race to discover the molecular structure of DNA *Modern Analytical Chemistry* David Harvey, 2000 This introductory text covers both traditional and contemporary topics relevant to analytical chemistry Its flexible approach allows instructors to choose their favourite topics of discussion from additional coverage of subjects such as sampling kinetic method and quality assurance **A Book on C** Al Kelley, Ira Pohl, 1990 The authors provide clear examples and thorough explanations of every feature in the C language They teach C vis a vis the UNIX operating system A reference and tutorial to the C programming language Annotation copyrighted by Book News Inc Portland OR **AP Chemistry For Dummies** Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13 A practical and hands on guide for learning the practical science of AP chemistry and preparing for the AP chem exam Gearing up for the AP Chemistry exam AP Chemistry For Dummies is packed with all the resources and help you need to do your very best Focused on the chemistry concepts and problems the College Board wants you to know this AP Chemistry study guide gives you winning test taking tips multiple choice strategies and topic guidelines as well as great advice on optimizing your study time and hitting the top of your game on test day This user friendly guide helps you prepare without perspiration by developing a pre test plan organizing your study time and getting the most out of your AP course You'll get help understanding atomic structure and bonding grasping atomic geometry understanding how colliding particles produce states and so much more To provide students with hands on experience AP chemistry courses include extensive labwork as part of the standard curriculum This is why the book dedicates a chapter to providing a brief review of common laboratory equipment and techniques and another to a complete survey of recommended AP chemistry experiments Two full length practice exams help you build your confidence get comfortable with test formats identify your strengths and weaknesses and focus your studies You'll discover how to Create and follow a pretest plan Understand everything you must know about the exam Develop a multiple choice strategy Figure out displacement combustion and acid base reactions Get familiar with stoichiometry Describe patterns and predict properties Get a handle on organic chemistry nomenclature Know your way

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Vector-Borne Diseases Institute of Medicine,Board on Global Health,Forum on Microbial Threats,2008-03-18 Vector borne infectious diseases such as malaria dengue fever yellow fever and plague cause a significant fraction of the global infectious disease burden indeed nearly half of the world s population is infected with at least one type of vector borne pathogen CIESIN 2007 WHO 2004a Vector borne plant and animal diseases including several newly recognized pathogens reduce agricultural productivity and disrupt ecosystems throughout the world These diseases profoundly restrict socioeconomic status and development in countries with the highest rates of infection many of which are located in the tropics and subtropics Although this workshop summary provides an account of the individual presentations it also reflects an important aspect of the Forum philosophy The workshop functions as a dialogue among representatives from different sectors and allows them to present their beliefs about which areas may merit further attention These proceedings summarize only the statements of participants in the workshop and are not intended to be an exhaustive exploration of the subject matter or a representation of consensus evaluation

Vector Borne Diseases Understanding the Environmental Human Health and Ecological Connections Workshop Summary Forum on Microbial Threats summarizes this workshop

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