


Solutions Chemistry WebQuest

Directions:

- Go to <http://www.wwnorton.com/college/chemistry/gilbert2/contents/ch04/studyplan.asp>
- Under Chapter 4 ChemTours click the link that matches with the following headings.
- Read through the text on each page and watch the visuals shown. When finished, click the play  button. If it asks you to click on anything, follow the instructions. Click on any *Concept Questions*, *Real-World Connections*, or *Science Connections* to get more information.

Molarity

1. Understanding the chemistry of liquid mixtures requires knowledge of amount or _____ of the compounds in the solution.
2. The concentration of solute is most usefully given in _____ (_____)
3. What is the equation for this?
4. How can you determine the number of moles of something?
5. Complete questions 1-7. Record your answers.
6. Calculating the molar concentration, or _____, of a solution is probably the _____ calculation done by _____ students and researchers.

Dilutions

7. Dilution occurs when additional _____ is added to the solution.
8. The _____ molecules are chemically unchanged during dilution.
9. When might you use a dilution? (read-world connections)
10. What is the equation used to solve dilution problems?
11. Both sides of the equation have exactly the same # of _____ of solute.
12. For solutions of the same compound in the same type of container, the _____ of the color is an indicator of solute _____.
13. The palest solution is the _____.
14. Answer questions 1& 2
15. Give 2 examples of when dilutions are used in science.

Migration of Ions

16. Some aqueous solutions will conduct _____ and some will not.
17. The flow of electricity is increased greatly when _____ ions travel to terminals & aid in _____ transfer.
18. _____ water is a poor conductor, but _____ or _____ water may contain ions that may cause electricity.
19. Why should you not go swimming outside during a thunderstorm?
20. In _____, ions can move freely and thereby conduct electricity.
21. What happens when a voltage is applied to Sodium Chloride Solution?
22. Why was the light bulb faint from the Acetic Acid solution?
23. Strong acid & bases, and ionic compounds are _____.
24. _____ & _____ are nonelectrolytes.
25. What types of ions would be found around the positive electrode?
26. What happens when 2 weak electrolytes are mixed?
27. Conductivity is another way to measure _____ in solution

Saturated Solutions

28. A saturated solution is one that contains the _____ concentration of _____.
29. A _____ is created when a substance is _____ in a _____.
30. When a solvent has reached its capacity and can dissolve no additional solute, it is said to be _____.
31. Give an example of a saturated solution.
32. What happens to compounds in an unsaturated solution?
33. In a saturated solution, excess molecules do not dissolve, but form a _____ on the bottom of the beaker.
34. What is happening to the molecule/ions in the beaker (section 3 of 4)?
35. What happens to solubility as temperature increases?

Solutions Chemistry Webquest Answers

James F Hall



Solutions Chemistry Webquest Answers

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real world is Real chemists use math and writing skills every day Their knowledge about how the scientific enterprise works is greatly enhanced by their knowledge of great scientists who came before Our approach to implementing a Kingdom Perspective does not entail unnaturally inserting Bible verses or devotional insets A better approach for Christians is to do science from the faith informed perspective of disciples of Christ studying God s created order This is apparent in General Chemistry from the observations about order in the universe in how nature submits to scientific study and modeling and in the fitness of creation as a habitat for humanity and animal life

Student's Selected Solutions Manual for Introductory Chemistry Nivaldo J. Tro, Matthew Johll, University of Illinois at Chicago, 2014-01-15 Student s Selected Solutions Manual by Matthew Johll of Illinois Valley Community College 9780321949073 0321949072 The selected solution manual for students contains complete step by step solutions to selected odd numbered end of chapter problems

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polymers are analyzed renders the text particularly topical Subjects discussed include Real ideal Gaussian semirigid and branched polymer chains Polymer solutions and thermodynamics Static light scattering of a polymer solution Dynamic light scattering and diffusion of polymers Dynamics of dilute and semidilute polymer solutions Study questions at the end of each chapter not only provide students with the opportunity to test their understanding but also introduce topics relevant to polymer solutions not included in the main text With over 250 geometrical model diagrams Polymer Solutions is a necessary reference for students and for scientists pursuing a broader understanding of polymers [Saanjhi Saanjh \(Vol - 3\)/ ?????](#) [???? \(??? - 3\) JV Manisha / ?? ?? ?????](#),2023-03-09 Saanjhi Saanjh Vol 3 is a collection of 12 short stories The book talks about everyday emotions and relations that are a part of our lives The stories center around the elders of our society The collection of short stories is an attempt to showcase the thoughts and feelings from our elders point of view The book is an attempt for all to understand what goes on in the mind and hearts of our parents Things that seem mundane to us can be the focal point for the elderly When we understand the depth and point of view of how our elders think we relate to them better and help bridge the inter generational gap [Glencoe Algebra 1, Student Edition McGraw-Hill,2002-02](#) Glencoe Algebra 1 is a key program in our vertically aligned high school mathematics series developed to help all students achieve a better understanding of mathematics and improve their mathematics scores on today s high stakes assessments

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