

**Experiment 11C:
Qualitative Analysis**

Name: _____
Date: _____

Follow the procedures on Part I and Part II of page 100-101 of the student lab manual. Fill in the following tables using all the procedures.

Table 1

Reagent	11C-11 Qualitative Analysis of Group 1 Cations: Lab 100-101				
	NH_3	H_2S	H_2O_2	H_2O	Reference
Group 1 cations					
Group 2 cations					
Group 3 cations					
Group 4 cations					

Table 2

Reagent	11C-12 Qualitative Analysis of Unknown Sample				
	NH_3	H_2S	H_2O_2	H_2O	Reference
Group 1					
Group 2 cations					
Group 3 cations (unknown ions)					
Group 4 cations					
Group 5 cations (unknown ions) Group 6 cations (unknown ions) Group 7					

Chemistry 12 Lab Qualitative Analysis 16e Answers

Clyde R. Metz, Mary E. Castellion



Chemistry 12 Lab Qualitative Analysis 16e Answers

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General Chemistry with Qualitative Analysis, Second Edition Ralph H. Petrucci, Robert K. Wismer, 1987 **Qualitative**
Analysis As a Laboratory Basis for the Study of General Chemistry William Conger Morgan, 1906 *A Laboratory*
Manual of Qualitative Analysis for Chemistry 3 Ralph Warren Gelbach, Lyle Ithiel Gilbertson, 1931 **A Series of**
Laboratory Procedures in Qualitative Analysis for High School Chemistry Galen Leroy Weick, 1967 A Course in
Qualitative Chemical Analysis (Classic Reprint) Charles Baskerville, 2017-10-11 Excerpt from A Course in Qualitative
Chemical Analysis Detailed methods for the preparation of solutions of definite strength are given for the assistance of the
instructor The value of introducing preliminary experiments in a course in Qualitative Analysis is a mooted question We
believe that they should be restricted to those which are utilized in the schemes of separation Experience has shown us that
it is a good plan to have in the laboratory several sets of bottles containing solutions of known strength of the salts of the
various metals The students are encouraged to use these to verify any of the preliminary tests in case of doubt By comparing
in special cases the results obtained with known solutions with those obtained with the unknown a definite knowledge of that
particular reaction is fixed in the student's mind Students should be encouraged to use short cuts and the use of the
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page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections
that remain are intentionally left to preserve the state of such historical works *Practical Chemistry* William Augustus
Tilden, 2018-01-09 Excerpt from Practical Chemistry The Principles of Qualitative Analysis The second part is devoted to
methods of Analysis Here he is first instructed in the properties of a few of the constituents of common salts and is taught
methods for their separation or identification in presence of one another If he goes no further than this he will at least have
learnt what chemical analysis means He afterwards proceeds to extend his knowledge till at the end of the course he will be
able to analyse any mixture of ordinary inorganic substances The analytical part is to a great extent shorn of directions for
manipulation as well as of details which are not relevant to the immediate object of the experiment According to my
experience these are not only of no use but are a positive hindrance to the apprehension of the facts to be acquired
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