

1. Explain why solutions are classified as mixtures instead of compounds.
2. Solutions are \_\_\_\_\_ mixtures made up of very small particles that are actually molecules, \_\_\_\_\_, or \_\_\_\_\_.
3. Solutions are said to be in a \_\_\_\_\_ phase even though the components may have been in different phases before the solution was formed.
4. Pure gold is 24 carat. 14-carat gold contains 14 parts gold and 10 parts other metals. 14-carat gold is said to be a(n) \_\_\_\_\_, which is a type of \_\_\_\_\_ solution. An example of a gaseous solution is \_\_\_\_\_, which is made up mostly of \_\_\_\_\_ and nitrogen when dry. The most common solutions are \_\_\_\_\_ solutions.
5. Define miscible:
6. Define immiscible:
7. Because of the \_\_\_\_\_, you can see the light beams from car headlights in a fog.
8. Multiple choice: To increase the rate of solution of a solid in water,
  - a. increase the pressure over the water.
  - b. decrease the pressure over the water.
  - c. crush the particles of the solid.
  - d. chill the water.
9. If 15 grams of iodine are dissolved in 1000 mL of alcohol, the alcohol is the (solute, solvent) and the solution is said to be a(n) \_\_\_\_\_.
10. A substance that dissolves other materials is a (solute, solvent). The substance being dissolved is a (solute, solvent).

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**Lei Shi**



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