



# Basic Chemistry

According to the arrangement of arrangement of atoms in space molecules that contain bonding between all atoms, although some are chemical substances that only have all their bonds in, for example, crystals, and arranged rather far from all of being correct.

The molecule contains in this chapter, however, from concepts of both, organic and inorganic chemistry. Chemistry is the science that studies the composition of matter, its physical properties, the chemical composition of matter, and the reactions that take place in it. It is a science that studies the relationship between the physical and chemical properties of matter, and the laws that govern them.

Understanding matter, structure, bonding, reactions of elements, and the chemical and physical of the most abundant compounds (ammonia, water, carbon dioxide, and oxygen) is a good starting point for studying other sciences in chemistry as well as other areas of the sciences.

## CONCEPTS OF MATTER AND ENERGY

1. Select all the correct statements that apply to each of the following processes and mark the letter in the correct column.

- |   |                                      |
|---|--------------------------------------|
| _____ 1. The energy needed to the bonds of <b>ammonia</b> : |                                      |
| a. is in the chemical bonds                                 | c. is in the chemical bonds          |
| b. is in the physical bonds                                 | d. is in the physical bonds          |
| _____ 2. Heat is:   |                                      |
| a. thermal energy   | c. thermal energy                    |
| b. electrical energy  | d. electrical energy                 |
| _____ 3. Molecules of water in liquid state:                |                                      |
| a. are in the state of free movement                        | c. are in the state of free movement |
| b. are in the state of free movement                        | d. are in the state of free movement |



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**L Darling-Hammond**



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