

100 Quick Fire CI Revision Questions

This worksheet is fully supported by a video tutorial

1. Where are protons?
2. Where are neutrons?
3. Where are electrons?
4. What is the charge on a proton?
5. What is the charge on a neutron?
6. What is the charge on an electron?
7. What is a compound?
8. What is a mixture?
9. What is an element?
10. How can we use the periodic table to find the number of protons an element has?
11. How can we use the periodic table to find the number of neutrons an element has?
12. How can we use the periodic table to find the number of electrons an element has?
13. How many electrons are in the first electron shell?
14. How many electrons are in the second electron shell?
15. How many electrons are in the third electron shell?
16. What do the groups in the periodic table tell us?
17. What do the periods on the periodic table tell us?
18. What type of ions (positive/negative) do metals form?
19. What type of ions (positive/negative) do non-metals form?
20. Balance this $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$
21. Balance this $\text{N}_2 + \text{H}_2 \rightarrow \text{NH}_3$
22. Balance this $\text{N}_2 + \text{Cl}_2 \rightarrow \text{NCl}_3$
23. How many elements in MgSO_4 ?
24. How many atoms in MgSO_4 ?
25. How many elements in $\text{Pb}(\text{OH})_2$?
26. How many atoms in $\text{Pb}(\text{OH})_2$?
27. What is the chemical name for limestone?
28. What is thermal decomposition?
29. What are the two products when CaCO_3 is thermally decomposed?
30. How do we test for carbon dioxide?
31. When we react a carbonate with a dilute acid what happens?
32. What we heat a metal carbonate what happens?
33. What we add water to calcium oxide what happens?
34. Give 3 advantages to lime stone quarries
35. Give 3 disadvantages to lime stone quarries

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