

2

Basic Chemistry

Knowledge in management is composed of both technical (hard) and social (soft) skills. Both types of skills are important, although some (or several) technical skills may be more critical than soft skills. Soft skills, such as communication, listening, and managing people are more likely to be being assessed.

The business systems in this chapter describe basic concepts of both business and people necessary. Concepts in the business-class include the composition of matter, atomic structure, chemical bonding, the physical composition of matter, solutions, acids, bases, and chemical reactions. Concepts in business include the relationship between the two disciplines of being organized, whether they are experiments, data, or research.

Understanding people, business, and organizational systems and the interaction and interface of the two is essential for organizational performance. This includes: skills and soft skills such as research methods, team building, ethics, processes in working to understand how the skills function.

CONCEPTS OF MATTER AND ENERGY

1. Select all answers that apply, parts of the following processes and their products include the formation of matter and energy.

- _____ 1. The energy stored in the bonds of hydrocarbons.
 - A. in hydrocarbons
 - B. in the intermolecular forces
 - C. in intermolecular distances
 - D. in the molecular structure
- _____ 2. Heat is:
 - A. thermal energy
 - B. thermal contact
 - C. kinetic energy
 - D. potential energy
- _____ 3. Molecular bonds in hydrocarbons:
 - A. are a form of energy
 - B. store energy in their bonds
 - C. release energy on bond
 - D. store energy on compound



Anatomy Chapter 2 Basic Chemistry Packet Answer Key

AN Whitehead



Anatomy Chapter 2 Basic Chemistry Packet Answer Key

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alcohols ethanol 2 methylpropan 2 ol butan 2 ol but 2 en 1 ol ch₃ch₂ch₂ch₂oh ii ch₃ ch₃ch₂ch₂oh iii ch₃ch₂ch₂ch₂oh iv ch₃ ch₃ch₂ch₂oh ch₃ 6 aldehydes and ketones draw the structures for the following compounds ethanal butanone also known as butan 2 one but the 2 is unnecessary and is often left out 2 methylpropanal

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question 1 basic hydrocarbon nomenclature part a enter the number of carbons indicated by each of the following root names in the designated answer box part b the line formula for a branched alkane is shown below part c the line formula for another branched alkane is

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jan 21 2020 the point of attachment is number one and any other smaller groups are named as substituent groups on the larger group this numbering is independent of the numbering of the parent chain try to name the following compounds

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naming organic compounds 1 1 alkanes a i you shouldn't really have got this one wrong unless you made a careless mistake like putting too many or too few hydrogens on the middle carbon if you got it wrong re read the bit about naming alkanes and then look at these questions again ii

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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with

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