

Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, using the letters in each "box" to identify the required element(s).

A																	B
												D	E				
C												H		K			
	F												I				
J	L	Z				M				N				O	P		
Q		R				S	T	U									

V											W				
X											Y				

List elements in alphabetical order when more than 1 element answers a question.

- _____ 1) Element **A** is in a box all by itself because: **A)** it is really a Noble Gas **B)** it is a gas **C)** it is NOT a gas **D)** it does NOT share the properties of the other elements in its group **E)** it is a liquid

- _____ 2) Which of these elements is/are highly reactive? **A)** "C" **B)** "Z" **C)** "K" **D)** A, B, and C **E)** A and B only

- _____ 3) What is the letter of the element whose atomic number is closest to the element labeled "Z"?

- _____ 4) Which letter labels the halogen element with the lowest atomic number?

- _____ 5) Which of these elements is the least reactive? **A)** "J" **B)** "Y" **C)** "A" **D)** "O" **E)** "B"

- _____ 6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? **A)** "C" **B)** "N" **C)** "M" **D)** "W" **E)** "Y"

continued on other side

Atoms And Periodic Table Review Answer Key

RJ Shavelson



Atoms And Periodic Table Review Answer Key

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The Chemical Alphabet ,1850

Understanding the Periodic Table ,2021-06-09

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