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The fifth edition of General Chemistry continues the tradition of

presenting only the material that is essential for a one-year general chemistry course. It strikes a balance between theory and application by incorporating real-world examples; helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity; and developing problem-solving and critical thinking skills. Although the fifth edition incorporates many new features, such as macro to micro artwork, six new animations correlated to the text, and the addition of new hand-drawn worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr. Chang's concise-but-thorough approach will appeal to efficiency-minded instructors and value-conscious students. Excerpt from Experiments Arranged for Students in General Chemistry This little work is designed as a guide for beginners in chemistry. The authors have found the course of instruction as arranged in the following pages productive of good; its object is not to dispense with the supervision of an instructor, but rather to assist him. The present edition differs from its predecessors in that new experiments have been introduced, while others described in former editions have been modified in accordance with the experience obtained in the daily use of the book with large classes. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works. Note: this is the standalone book, if you want the book/access card order the ISBN below: 0321633644 / 9780321633644 General Chemistry: Atoms First and MasteringChemistry[®] with Pearson eText Student Access Kit Package * Package consists of 0321570138 / 9780321570130 MasteringChemistry with Pearson eText Student Access Kit 0321571630 / 9780321571632 General Chemistry: Atoms First Being a student himself, Michael Rosen knows exactly how it feels when trying to learn difficult concepts. He has extensive experience tutoring General Chemistry and understands where students need help the most. THE GUIDE TO SURVIVING GENERAL CHEMISTRY is a straight-to-the-point resource that gives students exactly what they need to know, along with the best techniques for learning general chemistry. Endorsed by Professor Mark Oram, Ph.D for accuracy and topic coverage, this text is filled with study tips, example problems,

and worked out solutions for all topics in the course. Michael has written the ideal guide for mastering General Chemistry. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. 24 Laboratory exercises for general chemistry; On-line access to MSDS, Chemical literature, chemicals, equipment, glassware. This text includes the narrative from the MindTap General Chemistry Course. Important Notice: Media content referenced within the product description or the product text may not be available in the ebook version. Silberberg's Principles of General Chemistry offers students the same authoritative topic coverage as its parent text, Chemistry: The Molecular Nature of Matter and Change. The Principles text allows for succinct coverage of content with minimal emphasis on pedagogic learning aids. This more straightforward approach to learning appeals to today's efficiency-minded, value-conscious instructors and students without sacrificing depth, clarity, or rigor. The Survival Guide to Organic Chemistry: Bridging the Gap from General Chemistry enables organic chemistry students to bridge the gap between general chemistry and organic chemistry. It makes sense of the myriad of in-depth concepts of organic chemistry, without overwhelming them in the necessary detail often given in a complete organic chemistry text. Here, the topics covered span the entire standard organic chemistry curriculum. The authors describe subjects which require further explanation, offer alternate viewpoints for understanding and provide hands-on practical problems and solutions to help master the material. This text ultimately allows students to apply key ideas from their general chemistry curriculum to key concepts in organic chemistry. A question/answer supplement to accompany a 1-year general chemistry course. This text is a multiple choice format and can be used with any standard general chemistry text. The exercises emphasize the importance of problem-solving and equation writing in the style used in general chemistry examinations and professional school aptitude examinations such as the MCAT and VCAT. This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding. Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid/base reactions which often persists through

working with acid/base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshman in STEM fields. For two-semester courses in General Chemistry. Hill and Petrucci, two highly successful chemistry authors, take an exciting integrated approach to the concepts and applications of general chemistry. General Chemistry provides integrated coverage of organic and biochemistry; integrated applications; integrated tools that foster operational problem-solving skills and conceptual understanding; and an integrated media learning program. More than any other, this text offers balance in the topics presented, in its approach to problem solving, and in its presentation of the subject of chemistry. Equal emphasis is placed on both conceptual and quantitative problem solving. The Second Edition works to make chemistry more understandable to the average student, and features new and expanded coverage of key chemistry topics such as organic chemistry, biochemistry, material science, and environmental chemistry. More problems have been added, including illustrated problems and molecular models. Revised third edition of classic first-year text by Nobel laureate. Covers atomic and molecular structure, quantum mechanics, statistical mechanics, and thermodynamics correlated with descriptive chemistry. Problems. Understanding General Chemistry details the fundamentals of general chemistry through a wide range of topics, relating the structure of atoms and molecules to the properties of matter. Written in an easy-to-understand format with helpful pedagogy to fuel learning, the book features main objectives at the beginning of each chapter, get smart sections, and check your reading section at the end of each chapter. The text is filled with examples and practices that illustrate the concepts at hand. In addition, a summary, and extensive MCQs, exercises and problems with the corresponding answers and explanations are readily available. Additional features include: Alerts students to common mistakes and explains in simple ways and clear applications how to avoid these mistakes. Offers answers and comments alongside sample problems enabling students to self-evaluate their skill level. Includes powerful methods, easy steps, simple and accurate interpretations, and engaging applications to help students understand complex principles. Provides a bridge to more complex topics such as solid-state chemistry, organometallic chemistry, chemistry of main group elements, inorganic chemistry, and physical chemistry. This introductory textbook is ideal for chemistry courses for non-science majors as well as health sciences and preparatory engineering students. Emphasises on contemporary applications and an intuitive problem-solving approach that helps students discover the exciting potential of chemical science. This book incorporates fresh applications from the three major areas of modern research: materials, environmental chemistry, and biological science. This work has been selected by scholars as being culturally important and is part of the knowledge base of civilization as we know it. This work is in the public domain in the United States of America, and possibly other nations. Within the United States, you may freely copy and distribute this work, as no entity (individual or corporate) has a

copyright on the body of the work. Scholars believe, and we concur, that this work is important enough to be preserved, reproduced, and made generally available to the public. To ensure a quality reading experience, this work has been proofread and republished using a format that seamlessly blends the original graphical elements with text in an easy-to-read typeface. We appreciate your support of the preservation process, and thank you for being an important part of keeping this knowledge alive and relevant. Previous ed published: 1989 Periodic table and text on lining papers. Includes index and appendices. "General Chemistry for Engineers" is tailored for a one-semester freshman-level college course for students pursuing engineering degrees. The book offers a balance of conciseness, rigor, and depth needed to prepare students for more advanced coursework and careers in various engineering specialties, such as civil, environmental, electrical, computer, mechanical and industrial engineering, in addition to chemical engineering. This text leads students through the breadth of a typical two-semester sequence in general chemistry. It elucidates the key concepts and skills important for entering engineering students, including problem solving, qualitative and quantitative thinking, and importance of units. Examples are drawn from problems of interest to modern engineers, including alternative energy, advanced materials, and the environment. The book is the result of the author's unique experiences teaching approximately 2,500 freshman in chemistry and upper-level students in chemical and biological engineering, in addition to leading research and development teaching in the medical device and specialty pharmaceutical industries. The author received a variety of teaching awards at Northeastern honoring his work in making an intense, fast-pace course manageable and exciting. Paul A. DiMilla is an Associate Academic Specialist in Chemistry & Chemical Biology and Chemical Engineering at Northeastern University. He received his B.S. from the Massachusetts Institute of Technology and his Ph.D. from the University of Pennsylvania, both in Chemical Engineering. He was a Postdoctoral Fellow in Chemistry at Harvard University prior to beginning his faculty career in Chemical and Biomedical Engineering at Carnegie Mellon University, where he co-founded Automated Cell, Inc. Paul was a Visiting Professor of Bioengineering at the Franklin W. Olin College of Engineering and a Visiting Scholar in Biomedical Engineering at Boston University. Additionally, he led R&D teams in the private sector, developing tissue-engineered medical products and drug-generating biodegradable polymers. He received an Early Career Development Award from the NSF, a Searle Scholar Award, and the first Whitaker Young Investigator Award. He is also the inventor on seven issued US patents. All answers to exercises, quizzes and tests, plus worked out solutions. General Chemistry uses a mastery-learning method. In this method, students build comprehension by adding new concepts while reviewing and rehearsing key material throughout the year. This method is implemented in carefully crafted exercises, quizzes, and the textbook narrative, and it facilitates learning, mastery and retention. Students appreciate the smaller profile and lighter weight of our books. This is

possible because of aspects of our Textbook Philosophy. When mastery is the goal, you must cull down the material to an amount that can reasonably be mastered in one year. Related subjects are integrated into the narrative. The history of modern chemistry, mathematics and technical communication is emphasized throughout. Integration protects a course from seeming compartmentalized from other subjects. That's certainly not how the real world is. Real chemists use math and writing skills every day. Their knowledge about how the scientific enterprise works is greatly enhanced by their knowledge of great scientists who came before. Our approach to implementing a Kingdom Perspective does not entail unnaturally inserting Bible verses or devotional insets. A better approach for Christians is to do science from the faith-informed perspective of disciples of Christ studying God's created order. This is apparent in General Chemistry from the observations about order in the universe, in how nature submits to scientific study and modeling, and in the fitness of creation as a habitat for humanity and animal life. The most trusted general chemistry text in Canada is back in a thoroughly revised 11th edition. General Chemistry: Principles and Modern Applications, is the most trusted book on the market recognized for its superior problems, lucid writing, and precision of argument and precise and detailed and treatment of the subject. The 11th edition offers enhanced hallmark features, new innovations and revised discussions that that respond to key market needs for detailed and modern treatment of organic chemistry, embracing the power of visual learning and conquering the challenges of effective problem solving and assessment. Note: You are purchasing a standalone product; MasteringChemistry does not come packaged with this content. Students, if interested in purchasing this title with MasteringChemistry, ask your instructor for the correct package ISBN and Course ID. Instructors, contact your Pearson representative for more information. If you would like to purchase both the physical text and MasteringChemistry, search for: 0134097327 / 9780134097329 General Chemistry: Principles and Modern Applications Plus MasteringChemistry with Pearson eText -- Access Card Package, 11/e Package consists of: 0132931281 / 9780132931281 General Chemistry: Principles and Modern Applications 0133387917 / 9780133387919 Study Card for General Chemistry: Principles and Modern Applications 0133387801 / 9780133387803 MasteringChemistry with Pearson eText -- Valuepack Access Card -- for General Chemistry: Principles and Modern Applications Designed to help students recognize their learning style; understand how to read, classify, and create a problem-solving list; and practice problem-solving skills, each chapter provides study objectives and a summary of the text, followed by sample problems with detailed solutions, as well as true/false questions and a self test, with all answers provided at the end of the chapter. Problem-solving is one of the most challenging aspects students encounter in general chemistry courses leading to frustration and failure. Consequently, many students become less motivated to take additional chemistry courses after the first year. This book deals with calculations in general chemistry and its primary goal is to prevent frustration by

providing students with innovative, intuitive, and systematic strategies to problem-solving in chemistry. The material addresses this issue by providing several sample problems with carefully explained step-by-step solutions for each concept. Key concepts, basic theories, and equations are provided and worked examples are selected to reflect possible ways problems could be presented to students. Get a better grade in General Chemistry! Even though General Chemistry may be challenging at times; with hard work and the right study tools, you can still get the grade you want. With David Klein's General Chemistry as a Second Language, you'll be able to better understand fundamental principles of chemistry, solve problems, and focus on what you need to know to succeed. Here's how you can get a better grade in General Chemistry: Understand the basic concepts: General Chemistry as a Second Language focuses on selected topics in General Chemistry to give you a solid foundation. By understanding these principles, you'll have a coherent framework that will help you better understand your course. Study more efficiently and effectively: General Chemistry as a Second Language provides time-saving study tips and problem-solving strategies that will help you succeed in the course. Improve your problem-solving skills: General Chemistry as a Second Language will help you develop the skills you need to solve a variety of problem types - even unfamiliar ones! EXPERIMENTS IN GENERAL CHEMISTRY, Fourth Edition, has been designed to stimulate curiosity and insight, and to clearly connect lecture and laboratory concepts and techniques. To accomplish this goal, an extensive effort has been made to develop experiments that maximize a discovery-oriented approach and minimize personal hazards and ecological impact. Like earlier editions, the use of chromates, barium, lead, mercury, and nickel salts has been avoided. The absence of these hazardous substances should minimize disposal problems and costs. This lab manual focuses not only on what happens during chemical reactions, but also helps students understand 'why' chemical reactions occur. The sequence of experiments has been refined to follow topics covered in most general chemistry textbooks. In addition, Murov has included a correlation chart that links the experiments in the manual to the corresponding chapter topics in several Brooks/Cole general chemistry titles. Each experiment--framed by pre-and post-laboratory exercises and concluding thought-provoking questions--helps to enhance students' conceptual understanding. Excerpt from A Course in General Chemistry In preparing this textbook in general chemistry the authors have been guided by a few simple principles. They have endeavored to write a book scientific in spirit and at the same time thoroughly teachable. It may be said at once that the text presents few novelties, either in arrangement or in method. While individual teachers can succeed better by adopting unusual and original modes of presentation, we appear to have reached a more or less standard method, which is, on the whole, best adapted to the greatest number. The most noticeable departure from this arrangement in the present text is the postponement of the halogen elements to a relatively late chapter. Several considerations of pedagogy have prompted this change: (1) The student needs some training in experimental

manipulation before attempting laboratory exercises with these active elements. (2) The action of these elements upon the alkalis, and the conduct of their oxygen acids, involve many difficulties, and the attempt to meet these by presenting the chemistry of these elements in several widely separated chapters is bewildering to the student. (3) It is unsatisfactory to develop the conception of an acid by means of the example of the halogen hydrides, since they are far from being typical acids. In the experience of the authors the order followed in this text is much to be preferred. In the main the elements have been grouped in description in accordance with their positions in the periodic classification, but the authors have not hesitated to abandon this order when a more natural one suggested itself. Thus, with the metals the arrangement chosen presents the strongly electropositive elements of unchanging valence first. These are followed by those which present the complications of changeable valence, sulfide metallurgy, and amphoteric characteristics. Every teacher desires to impress his students with the truth that science rests fundamentally upon experiment and is therefore subject to the uncertainties of experimental error. The authors believe that this end is best attained by the free introduction of historical matter. About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books. Find more at www.forgottenbooks.com This book is a reproduction of an important historical work. Forgotten Books uses state-of-the-art technology to digitally reconstruct the work, preserving the original format whilst repairing imperfections present in the aged copy. In rare cases, an imperfection in the original, such as a blemish or missing page, may be replicated in our edition. We do, however, repair the vast majority of imperfections successfully; any imperfections that remain are intentionally left to preserve the state of such historical works. 23 Laboratory exercises and associated prelaboratory material for general chemistry; On-line access to MSDS, Chemical literature, chemicals, equipment, glassware. Integrates descriptive chemistry with chemical principles to allow students to learn difficult concepts. The chapter on qualitative analysis deals with all the major types of reactions encountered in chemistry. The sixth edition of General Chemistry continues the tradition of presenting only the material that is essential for a one-year general chemistry course. It strikes a balance between theory and application by incorporating real-world examples; helping students visualize the three-dimensional atomic and molecular structures that are the basis of chemical activity; and developing problem-solving and critical thinking skills. Although the sixth edition incorporates many impressive features, such as macro to micro artwork, animations correlated to the text, and hand-sketched worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr. Chang's concise-but-thorough approach will appeal to efficiency-minded instructors and value-conscious students. The leading lab manual for general chemistry courses In the newly refreshed eleventh edition of Laboratory Manual for Principles of General Chemistry, dedicated researchers Mark Lassiter and J. A. Beran deliver an essential manual perfect for students seeking a wide variety of

experiments in an easy-to understand and very accessible format. The book contains enough experiments for up to three terms of complete instruction and emphasizes crucial chemical techniques and principles. This best-selling text is summarized by "classic text, modern presentation." Its emphasis on fundamental skills and concepts and clearly explained problem-solving strategies continues to be a strength. The revision builds on the highly praised style and applications to everyday life that have earned this text a reputation as the voice of authority in general chemistry. THE MAIN OBJECTIVES DURING THE DEVELOPMENT OF THIS BOOK WAS TO BETTER PREPARE STUDENTS THAT NEED TO TAKE THE FINAL COMPREHENSIVE AMERICAN CHEMICAL SOCIETY (ACS) EXAM, AS WELL AS THOSE STUDENTS THAT SEEK ADMISSION IN MEDICAL AND PHARMACY SCHOOLS. THE STUDY GUIDE DESCRIBES IN AN OUTLINE FORMAT THE MOST IMPORTANT TOPICS COVERED IN GENERAL CHEMISTRY I. THE BOOK SUMMARIZES THE MAIN OBJECTIVES THAT STUDENTS ENROLLED IN THE COURSE SHOULD LEARN. THE BOOK GIVES MANY SAMPLE PROBLEMS , WITH STEP WISE SOLUTIONS SO THAT STUDENTS CAN FOLLOW THE MATERIAL EASIER. WHEN NECESSARY, MATHEMATICAL FORMULAS ARE GIVEN ALL THROUGHOUT TO FACILITATE THE SOLUTIONS TO NUMERICAL PROBLEMS. THIS STUDY GUIDE CAN BE USED BY ANY COLLEGE STUDENT ENROLLED IN GEN CHEM I, REGARDLESS OF THE TEXT

USED. ONE OF THE PITFALLS OF MOST TEXTS IS THE EXCESSIVE AMOUNT OF MATERIAL COVERED, BUT FAIL TO EMPHASIZE THE MOST IMPORTANT FEATURES OF THE TOPIC COVERED.

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