

Oxidation-Reduction Worksheet 1

1. Identify the following in the reaction: $2\overset{+2}{\text{Fe}}^{2+} + \overset{0}{\text{Cl}}_2 \rightarrow 2\overset{+3}{\text{Fe}}^{3+} + 2\overset{-1}{\text{Cl}}^{-}$ (+ marks)

a. The oxidizing agent	Cl_2	e. The species gaining electrons	Cl_2
b. The species being oxidized	Fe^{2+}	f. The species losing electrons	Fe^{2+}
c. The reducing agent	Fe^{2+}	g. The product of oxidation	Fe^{3+}
d. The species being reduced	Cl_2	h. The product of reduction	Cl^{-}

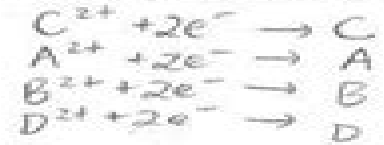
2. Use the Standard Reduction Table to answer the following. (+ marks)

- Will Br_2 oxidize $\text{Au}_{(s)}$? Explain. Yes, $\text{Br}_{2(s)}$ is higher on the table as an oxidizing agent than $\text{Au}_{(s)}$ as a reducing agent
- Give the symbol of an ion that will oxidize $\text{Mn}_{(s)}$ but not $\text{Cr}_{(s)}$. Zn^{2+}
- Which is a stronger reducing agent, Sn^{2+} or Fe^{2+} ? Sn^{2+}
- Can you safely put a gold ring in acidified dichromate solution? What about acidified bromate solution? Explain. $\text{Au}_{(s)}$ is higher on the table as a reducing agent than $\text{Cr}_2\text{O}_7^{2-} + 14\text{H}^+$ as an oxidizing agent, so it will not react. $\text{Au}_{(s)}$ will not react with $\text{BrO}_3^- + 6\text{H}^+$ either for same reason

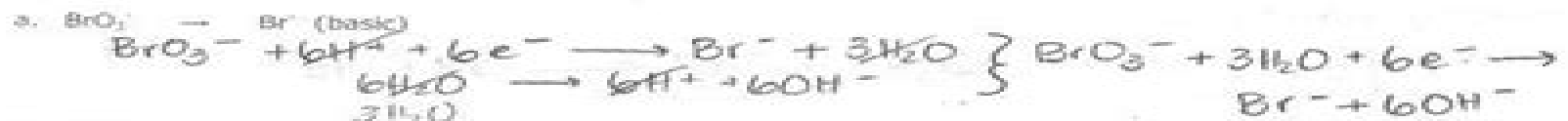
3. Four metals A, B, C, & D were tested with separate solutions of A^{2+} , B^{2+} , C^{2+} & D^{2+} . Some of the results are summarized in the following table:

Metal	Solution			
	A^{2+}	B^{2+}	C^{2+}	D^{2+}
A		(i) no reaction	(ii) reaction	
B				(iii) no reaction
D	(iv) reaction			

List the ions in order from the strongest to weakest oxidizing agent. (2 marks)



4. Balance the following half reactions. (3 marks)



Prentice Hall Oxidation And Reduction Answer Key

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Inorganic Chemistry in Aqueous Solution Jack Barrett, 2003 Inorganic Chemistry in Aqueous Solution reviews the chemistry of the elements in all their oxidation states in an aqueous environment The nature of ions in solution is described in some detail and enthalpies and entropies of hydration of many ions are defined and recalculated from the best data available These values are used to provide an understanding of the periodicities of standard reduction potentials Standard reduction potential data for all of the elements group by group covering the s and p d and f blocks of the Periodic Table is also included Major sections are devoted to the acid base behaviour and the solubilities of inorganic compounds in water Inorganic Chemistry in Aqueous Solution is aimed at undergraduate chemistry students but will also be welcomed by geologists interested in this field Ideal for the needs of undergraduate chemistry students Tutorial Chemistry Texts is a major series consisting of short single topic or modular texts concentrating on the fundamental areas of chemistry taught in undergraduate science courses Each book provides a concise account of the basic principles underlying a given subject embodying an independent learning philosophy and including worked examples Standard Potentials in Aqueous Solution Allen J. Bard, 2017-11-22 The best available collection of thermodynamic data The first of its kind in over thirty years this up to date book presents the current knowledge on Standard Potentials in Aqueous Solution Written by leading international experts and initiated by the IUPAC Commissions on Electrochemistry and Electroanalytical Chemistry this remarkable work begins with a thorough review of basic concepts and methods for determining standard electrode potentials Building upon this solid foundation this convenient source proceeds to discuss the various redox couples for every known element The chapters of this practical time saving guide are organized in order of the groups of elements on the periodic table for easy reference to vital material AND each chapter also contains the fundamental chemistry of elements numerous equations of chemical reactions easy to read tables of thermodynamic data and useful oxidation state diagrams Standard Potentials in Aqueous Solution is an ideal handy reference for analytical and physical chemists electrochemists electroanalytical chemists chemical engineers biochemists inorganic and organic chemists and spectroscopists needing information on reactions and thermodynamic data in inorganic chemistry And it is a valuable supplementary text for undergraduate and graduate level chemistry students **Oxidizing and Reducing Agents** Steven D. Burke, Rick L. Danheiser, 1999-07-09 Oxidizing and Reducing Agents S D Burke University of Wisconsin at Madison USA R L Danheiser Massachusetts Institute of Technology Cambridge USA Recognising the critical need for bringing a handy reference work that deals with the most popular reagents in synthesis to the laboratory of practising organic chemists the Editors of the acclaimed Encyclopedia of Reagents for Organic Synthesis EROS have selected the most important and useful reagents employed in contemporary organic synthesis Handbook of Reagents for Organic Synthesis Oxidizing and Reducing Agents provides the synthetic chemist with a

convenient compendium of information concentrating on the most important and frequently employed reagents for the oxidation and reduction of organic compounds extracted and updated from EROS The inclusion of a bibliography of reviews and monographs a compilation of Organic Syntheses procedures with tested experimental details and references to oxidizing and reducing agents will ensure that this handbook is both comprehensive and convenient Addison-Wesley Chemistry Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, Prentice-Hall Staff, 2001-02 To purchase or download a workbook click on the Purchase or Download button to the left To purchase a workbook enter the desired quantity and click Add to Cart To download a free workbook right click the FREE Download PDF link and save to your computer This will result in a faster download as opposed to left clicking and opening the link **Assessment that Informs Practice**, 2000 **Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science**, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *JEE Main 2020 Chemistry - Unit wise Practice Test Papers* Career Point Kota, 2020-07-19 Competitive examination preparation takes enormous efforts time on the part of a student to learn practice and master each unit of the syllabus To check proficiency level in each unit student must take self assessment to identify his her weak areas to work upon that eventually builds confidence to win Also performance of a student in exam improves significantly if student is familiar with the exact nature type and difficulty level of the questions being asked in the Exam With this objective in mind we are presenting before you this book containing unit tests Some features of the books are The complete syllabus is divided into logical units and there is a self assessment tests for each unit Tests are prepared by subject experts who have decade of experience to prepare students for competitive exams Tests are as per the latest pattern of the examination Detailed explanatory solution of each test paper is also given Student is advised to attempt these Tests once they complete the preparation revision of unit They should attempt these Test in exam like environment in a specified time Student is advised to properly analyze the solutions and think of alternative methods and linkage to the solutions of identical problems also We firmly believe that the book in this form will definitely help a genuine hardworking student We have put our best efforts to make this book error free still there may be some errors We would appreciate if the same is brought to our notice We wish to utilize the opportunity to place on record our special thanks to all faculty members and editorial team for their efforts to make this book *Report of Investigations*, 1953 **Innovative Curriculum Materials**, 1999 **Chemistry 2e** Paul Flowers, Richard Langely, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and

understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition **Certain Aspects of the Galvanic Corrosion Behavior of Titanium** David Schlain,1953

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Prentice Hall Chemistry Antony C. Wilbraham,2006-10-15 Prentice Hall Chemistry meets the needs of students with a range of abilities diversities and learning styles by providing real world connections to chemical concepts and processes The first nine chapters introduce students to the conceptual nature of chemistry before they encounter the more rigorous mathematical models and concepts in later chapters The technology backbone of the program is the widely praised Interactive Textbook with ChemASAP which provides frequent opportunities to practice and reinforce key concepts with tutorials that bring chemistry to students through Animations Simulations Assessment and Problem solving tutorials

Library of Congress Catalog: Motion Pictures and Filmstrips Library of Congress,1968 **ENC Focus** ,2000

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Jean-Claude Bertrand,Pierre Caumette,Philippe Lebaron,Robert Matheron,Philippe Normand,Télesphore

Sime-Ngando,2015-01-26 This book is a treatise on microbial ecology that covers traditional and cutting edge issues in the ecology of microbes in the biosphere It emphasizes on study tools microbial taxonomy and the fundamentals of microbial activities and interactions within their communities and environment as well as on the related food web dynamics and biogeochemical cycling The work exceeds the traditional domain of microbial ecology by revisiting the evolution of cellular prokaryotes and eukaryotes and stressing the general principles of ecology The overview of the topics authored by more than 80 specialists is one of the broadest in the field of environmental microbiology The overview of the topics authored by more than 80 specialists is one of the broadest in the field of environmental microbiology *Geological Survey Professional Paper* Geological Survey (U.S.),1966 *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office,1975

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


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