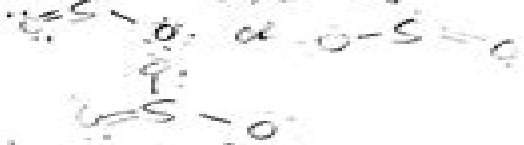
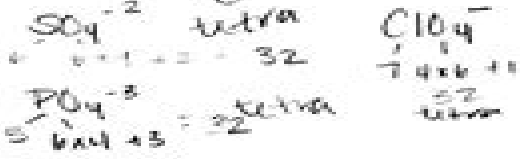


linear 2.0 sp
 2, 3 sp3d



hydrogen bonding



most polar

π - double bond words

$\text{P}=\text{O} = \text{X} \text{P}$
 $= (\frac{1}{2}) (780)$
 dipole-dipole vs London

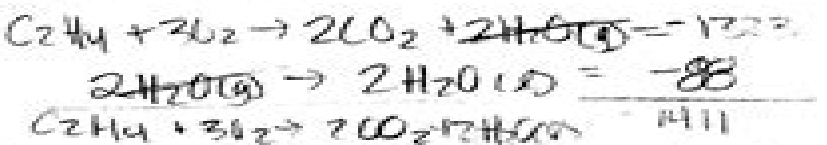
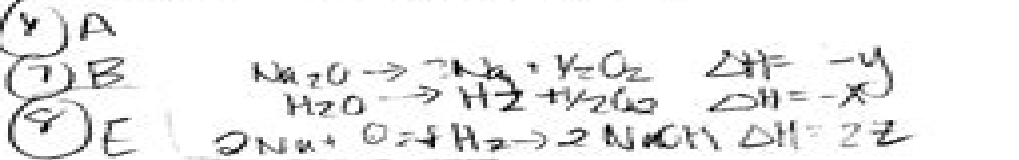
(6) A $\text{CW} + \text{X} \rightarrow \text{Y} + \text{Z}$
 (7) A $\begin{pmatrix} 1.2 & 1.6 \\ .2 & .7 \end{pmatrix} \begin{pmatrix} 0 \\ 0 \end{pmatrix} = \begin{pmatrix} 2 \\ 2 \end{pmatrix}$
 (8) C $\begin{pmatrix} 1.2 & 1.6 \\ .2 & .7 \end{pmatrix} \begin{pmatrix} 0 \\ 0 \end{pmatrix} = \begin{pmatrix} 2 \\ 2 \end{pmatrix}$
 All 1:1 so chem will see .2

$v_{rms} = \sqrt{\frac{3RT}{M}}$
 $PV = nRT$
 $P(D) = .09 (.0521) (600)$
 $P = 2.956$

(9) A the fastest - most gone

Therm
 (1) B Bond = \sum broken - \sum formed
 Enthalpy = $(150 + 3(248)) - (6(210))$

(2) C $S \rightarrow L$ EQ
 (3) D $S \rightarrow L \rightarrow G$ entropy
 (4) C $-889.1 = (-392.3 + x(228.5)) - x$
 $-889.1 = -965.1 - x$
 $x = -76$
 (5) E $S \rightarrow L$ entropy, endo, spont, + + -



Phase
 (1) E
 (2) A
 (3) E (Ag is too small to filter)
 (4) B
 (6) B
 (7) B
 (8) C
 (9) A
 (10) A

$\begin{matrix} \text{S/L} \\ \text{G} \end{matrix}$
 $\begin{matrix} 100 \\ 100 \end{matrix} \rightarrow \begin{matrix} 50 \\ 50 \end{matrix} \rightarrow \begin{matrix} 2 \\ 2 \end{matrix}$
 100 - 87.5 = 12.5
 100 - 50 = 50
 100 - 50 = 50

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This book takes a fresh look at programs for advanced studies for high school students in the United States with a particular focus on the Advanced Placement and the International Baccalaureate programs and asks how advanced studies can be significantly improved in general. It also examines two of the core issues surrounding these programs: they can have a profound impact on other components of the education system and participation in the programs has become key to admission at selective institutions of higher education. By looking at what could enhance the quality of high school advanced study programs as well as what precedes and comes after these programs, this report provides teachers, parents, curriculum developers, administrators, college science and mathematics faculty, and the educational research community with a detailed assessment that can be used to guide change within advanced study programs. *Conceptual Chemistry Volume I For Class XI* S.K. Jain & Shailesh K. Jain, 1998

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