



Name: _____
Date: _____

COUNTING ATOMS

WORKSHEET

- Chemical formulas can have three components.
 - The _____ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
 - The _____ which are numbers that are found on the lower _____ - hand side of each element symbol. This tells you the number of _____ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only _____ atom of this element in the molecule.
 - The _____ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be _____ by the _____ on the outside of the bracket found on the lower _____ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

2) HNO ₃	
Element	# of Atoms
Total	

3) Li ₂ O	
Element	# of Atoms
Total	

4) Li ₂ SO ₄	
Element	# of Atoms
Total	

5) NH ₄ Cl	
Element	# of Atoms
Total	

6) CaClO ₄	
Element	# of Atoms
Total	

7) H ₂ COCH ₂	
Element	# of Atoms
Total	

8) Mg(OH) ₂	
Element	# of Atoms
Total	

9) Al(OH) ₃	
Element	# of Atoms
Total	

10) NH ₄ C ₂ H ₃ O ₂	
Element	# of Atoms
Total	

11) NaC ₂ H ₃ O ₂	
Element	# of Atoms
Total	

12) (NH ₄) ₃ PO ₄	
Element	# of Atoms
Total	

13) Ca ₃ (PO ₄) ₂	
Element	# of Atoms
Total	

14) Fe ₂ (CO ₃) ₃	
Element	# of Atoms
Total	

15) Al ₂ (SO ₄) ₃	
Element	# of Atoms
Total	

Counting Atoms Answers

Lawrie Ryan



Counting Atoms Answers

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National 5 Chemistry with Answers: Second Edition Barry McBride, Stephen Jeffrey, John Anderson, 2018-07-30 Exam Board SQA Level National 5 Subject Chemistry First Teaching August 2017 First Exam May 2018 The second edition of this textbook has been fully revised and updated to reflect changes made to the SQA syllabus from 2017 onwards New features include Refreshed content Additional candidate advice Model answers for open ended questions

Investigating the Chemistry of Atoms Elizabeth Cregan, 2007-08-03 Atoms are the building blocks of everything around us Inside the atom is the nucleus with protons and neutrons But there are even smaller particles that make up protons and neutrons called quarks Everything that scientists know about atoms is part of the Atomic Theory which originated in ancient Greece but continues even today

Atoms, Molecules & Elements: The Periodic Table Gr. 5-8 George Graybill, 2015-10-01 This is the chapter slice The Periodic Table from the full lesson plan Atoms Molecules Elements Young scientists will be thrilled to explore the invisible world of atoms molecules and elements Our resource provides ready to use information and activities for remedial students using simplified language and vocabulary Students will label each part of the atom learn what compounds are and explore the patterns in the periodic table of elements to find calcium Ca chlorine Cl and helium He through hands on activities These and more science concepts are presented in a way that makes them more accessible to students and easier to understand Written to grade and using simplified language and vocabulary and comprised of reading passages student activities crossword word search comprehension quiz and color mini posters our resource can be used effectively for test prep and your whole class All of our content is aligned to your State Standards and are written to Bloom s Taxonomy and STEM initiatives

Paradigms of Artificial Intelligence Programming Peter Norvig, 2014-06-28 Paradigms of AI Programming is the first text to teach advanced Common Lisp techniques in the context of building major AI systems By reconstructing authentic complex AI programs using state of the art Common Lisp the book teaches students and professionals how to build and debug robust practical programs while demonstrating superior programming style and important AI concepts The author strongly emphasizes the practical performance issues involved in writing real working programs of significant size Chapters on troubleshooting and efficiency are included along with a discussion of the fundamentals of object oriented programming and a description of the main CLOS functions This volume is an excellent text for a course on AI programming a useful supplement for general AI courses and an indispensable reference for the professional programmer

Solutions Manual to Accompany Inorganic Chemistry 7th Edition Alen Hadzovic, 2018 This solutions manual accompanies the 7th edition of Inorganic chemistry by Mark Weller Tina Overton Jonathan Rourke and Fraser Armstrong As you master each chapter in Inorganic Chemistry having detailed solutions handy allows you to confirm your answers and develop your ability to think through the problem solving process

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CliffsNotes Chemistry Practice Pack Charles Henrickson,2010-02-08 Reviews chemistry topics with problems and solutions throughout and includes a customized adaptable full length exam

Chemistry 2e Paul Flowers,Richard Langely,William R. Robinson,Klaus Hellmut Theopold,2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition

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Basic Concepts of Chemistry, Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's *Basic Concepts of Chemistry* provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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