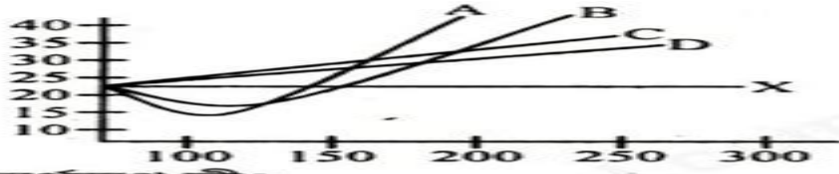
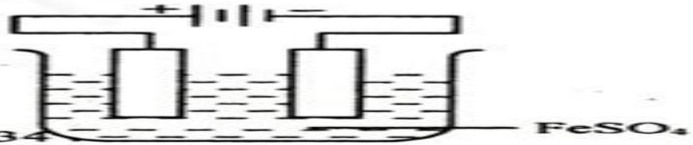


কর্তব্য :- প্রতিটি প্রশ্নে যথা প্রয়োজনীয় উত্তর দাও। প্রশ্ন উল্লেখিত মতে উত্তর দাও। প্রতিটি প্রশ্নের মোট উত্তর দাও। যে কোনো চারটি প্রশ্নের উত্তর দাও।

১. কয়েকটি অজানা গ্যাসের চাপ (P) ও আয়তন (V) নির্ণয় করে PV বনাম P লেখচিত্র অঙ্কন করে নিম্নলিখিত লেখ পাওয়া যায়।



- ক. বর্ণমূল গড় বর্ণবেগ কী? 1/2
 - খ. $AlCl_3$ কে অম্লধর্মী যৌগ বলা হয় কেন? 1/2
 - গ. X এর ন্যায় বৈশিষ্ট্যধারী গ্যাসের জন্য একটি সমীকরণ প্রতিপাদন কর। 1/2
 - ঘ. উদ্ভীপকের বিভিন্ন গ্যাসকে কি কি নামে অভিহিত করা যায়? A গ্যাস কি শর্তে X এর ন্যায় আচরণ করবে? 1/2
২. কয়েকটি ধাতুর প্রমাণ তড়িৎদ্বার বিভিন্ন এবং একটি কোষচিত্র নিম্নরূপ—



- (i) $E^{\circ}_{Cu/Cu^{2+}} = -0.34V$
- (ii) $E^{\circ}_{Fe/Fe^{2+}} = +0.44V$
- (iii) $E^{\circ}_{Zn/Zn^{2+}} = -0.76V$
- ক. হেটোরো অ্যারোমেটিক যৌগের উদাহরণ দাও। 1/2
- খ. অনুবন্দী অম্ল ও অনুবন্দী ক্ষারক ব্যাখ্যা কর। 1/2
- গ. 250 A বিদ্যুৎ 20 মিনিট চালনা করলে কি পরিমাণ ধাতু ক্যাথোডে জমা হবে? 1/2

- ঘ. উদ্ভীপকের দ্রবণটি কপার পত্র ও জিংক পাত্রে রাখা যাবে কিনা? ব্যাখ্যা কর। 1/2
- ৩. $KMnO_4 + H_2SO_4 + FeSO_4 \rightarrow$ 1/2
- ক. পেপটাইড বন্ধন কী? 1/2
- খ. কোন গ্যাস মিশ্রণের মোট চাপ, আংশিক চাপ ও মোল ভগ্নাংশের মধ্যে সম্পর্ক স্থাপন কর। 1/2
- গ. উদ্ভীপকের সমীকরণটি পূর্ণ কর এবং আয়ন-ইলেকট্রন পদ্ধতিতে সমতা বিধান কর। 1/2
- ঘ. উদ্ভীপকের জারকের, 0.1 M মাত্রার 20 ml দ্রবণ দ্বারা সমান মাত্রার কি পরিমাণ $FeSO_4$ দ্রবণকে জারিত করা যাবে? 1/2

- ৪. $CH_3 - CH_2 - CH_2 - Br \xrightarrow{NaOH(aq)} A$
- $\xrightarrow{NaOH(alc)} B$
- ক. কার্বক্সিল মূলকের গাঠনিক সংকেত লিখ। 1/2
- খ. $CH_3 - NH_2$ ও $CH_3 - CH_2 - NH_2$ এর মধ্যে কোনটি অম্লিক ক্ষারীয় এবং কেন? 1/2
- গ. সমীকরণ পূর্ণ করে A ও B এর মান লিখ। 1/2
- ঘ. A ও HBr বিক্রিয়া পূর্ণপূর্বক কোনটি প্রধান উৎপাদ? যুক্তিসহ ব্যাখ্যা কর। 1/2

- ৫. $\text{C}_6\text{H}_5\text{Cl} \xrightarrow{(i) NaOH(aq), 300^\circ C, 150 atm} A$
- $\xrightarrow{(ii) HCl(aq)} B$
- ক. ক্যান্টিনেশন কী? 1/2
- খ. অ্যারোমেটিসিটি কী? 1/2
- গ. আলকাতরা থেকে A যৌগটির প্রস্তুত বর্ণনা কর। 1/2
- ঘ. A যৌগের প্রকৃতি কারণসহ ব্যাখ্যা কর। 1/2

- ৬. $CH_3 - CH_2 - CH_2 - OH \xrightarrow{(i) [O]} C_3H_6O \rightarrow B$
- $\xrightarrow{(ii) Cu(OH)_2 + NaOH} \text{হলুদ অধঃক্ষেপ}$
- ক. BOD কী? 1/2
- খ. জৈব যৌগে কার্বক্সিল মূলক সনাক্তকরণের পরীক্ষা দাও। 1/2
- গ. উদ্ভীপকের আলোককে প্রয়োজনীয় সমীকরণসহ 'B' যৌগটি সনাক্ত কর। 1/2
- ঘ. সমান সংখ্যক কার্বন পরমাণু বিশিষ্ট একটি যৌগের সংকেত লিখ ও সমীকরণ দাও যা উদ্ভীপকের (i) ও (ii) নং বিকারকের সহিত বিক্রিয়া করে। 1/2

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Table of Contents Chemistry 2014 Hsc Objective Que 2nd Paper

1. Understanding the eBook Chemistry 2014 Hsc Objective Que 2nd Paper
 - The Rise of Digital Reading Chemistry 2014 Hsc Objective Que 2nd Paper
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry 2014 Hsc Objective Que 2nd Paper
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry 2014 Hsc Objective Que 2nd Paper
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry 2014 Hsc Objective Que 2nd Paper

- Personalized Recommendations
 - Chemistry 2014 Hsc Objective Que 2nd Paper User Reviews and Ratings
 - Chemistry 2014 Hsc Objective Que 2nd Paper and Bestseller Lists
5. Accessing Chemistry 2014 Hsc Objective Que 2nd Paper Free and Paid eBooks
- Chemistry 2014 Hsc Objective Que 2nd Paper Public Domain eBooks
 - Chemistry 2014 Hsc Objective Que 2nd Paper eBook Subscription Services
 - Chemistry 2014 Hsc Objective Que 2nd Paper Budget-Friendly Options
6. Navigating Chemistry 2014 Hsc Objective Que 2nd Paper eBook Formats
- ePub, PDF, MOBI, and More
 - Chemistry 2014 Hsc Objective Que 2nd Paper Compatibility with Devices
 - Chemistry 2014 Hsc Objective Que 2nd Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Chemistry 2014 Hsc Objective Que 2nd Paper
 - Highlighting and Note-Taking Chemistry 2014 Hsc Objective Que 2nd Paper
 - Interactive Elements Chemistry 2014 Hsc Objective Que 2nd Paper
8. Staying Engaged with Chemistry 2014 Hsc Objective Que 2nd Paper
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry 2014 Hsc Objective Que 2nd Paper
9. Balancing eBooks and Physical Books Chemistry 2014 Hsc Objective Que 2nd Paper
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry 2014 Hsc Objective Que 2nd Paper
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemistry 2014 Hsc Objective Que 2nd Paper
- Setting Reading Goals Chemistry 2014 Hsc Objective Que 2nd Paper
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemistry 2014 Hsc Objective Que 2nd Paper

- Fact-Checking eBook Content of Chemistry 2014 Hsc Objective Que 2nd Paper
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
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