

Types of Chemical Reactions Worksheet #1

H/Chemistry

For each of the following equations, identify what kind of reaction it represents: double replacement, single replacement, decomposition, or synthesis (composition).

- 1) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ **Synthesis**
- 2) $\text{CaI}_2 + \text{Cl}_2 \rightarrow \text{CaCl}_2 + \text{I}_2$ **Single Replacement**
- 3) $3\text{KOH} + \text{AlCl}_3 \rightarrow \text{Al(OH)}_3 + 3\text{KCl}$ **Double Replacement**
- 4) $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$ **Synthesis**
- *5) $\text{Ca(OH)}_2 + 2\text{HCl} \rightarrow \text{CaCl}_2 + 2\text{H}_2\text{O}$ **Double Replacement**
- 6) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ **Decomposition**
- 7) $3\text{CuSO}_4 + 2\text{Al} \rightarrow \text{Al}_2(\text{SO}_4)_3 + 3\text{Cu}$ **Single Replacement**
- 8) $\text{Na}_2\text{S} + 2\text{AgNO}_3 \rightarrow 2\text{NaNO}_3 + \text{Ag}_2\text{S}$ **Double Replacement**
- 9) $3\text{H}_2 + \text{N}_2 \rightarrow 2\text{NH}_3$ **Synthesis**
- *10) $2\text{Na} + 2\text{H}_2\text{O} \rightarrow 2\text{NaOH} + \text{H}_2$ **Single Replacement**

*Remember: "H₂O" can also be thought of as "HOH"

Complete each of the following equations as needed to make it the type of reaction indicated. Be sure to write each formula correctly.

- 11) Double replacement: $\text{Na}_2\text{CrO}_4 + \text{PbCl}_2 \rightarrow 2\text{NaCl} + \text{PbCrO}_4$
- 12) Single replacement: $\text{Cl}_2 + 2\text{NaBr} \rightarrow 2\text{NaCl} + \text{Br}_2$
- 13) Decomposition: $\text{Mg(ClO}_3)_2 \rightarrow \text{MgCl}_2 + 3\text{O}_2$
- 14) Synthesis: $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
- 15) Double replacement: $3\text{Ca(OH)}_2 + 2\text{FeCl}_3 \rightarrow 2\text{Fe(OH)}_3 + 3\text{CaCl}_2$
- 16) Single replacement: $\text{Fe} + \text{Cu(NO}_3)_2 \rightarrow \text{Fe(NO}_3)_2 + \text{Cu}$ [Assume Fe²⁺]
- 17) Decomposition: $2\text{Hg}_2\text{O} \rightarrow 4\text{Hg} + \text{O}_2$
- 18) Synthesis: $\text{S} + \text{O}_2 \rightarrow \text{SO}_2$
- 19) Double replacement: $\text{AgNO}_3 + \text{KI} \rightarrow \text{AgI} + \text{KNO}_3$
- 20) Single replacement: $\text{Cu} + 2\text{AgNO}_3 \rightarrow 2\text{Ag} + \text{Cu(NO}_3)_2$ [Copper (II) is used here]

Introduction To Chemistry Answer Key

Antony C. Wilbraham



Introduction To Chemistry Answer Key

Chemistry 2e Paul Flowers, Richard Langley, William R. Robinson, Klaus Hellmut Theopold, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition *An Introduction to Chemistry* Mark Bishop, 2002 This book teaches chemistry at an appropriate level of rigor while removing the confusion and insecurity that impair student success Students are frequently intimidated by prep chem Bishop's text shows them how to break the material down and master it The flexible order of topics allows unit conversions to be covered either early in the course as is traditionally done or later allowing for a much earlier than usual description of elements compounds and chemical reactions The text and superb illustrations provide a solid conceptual framework and address misconceptions The book helps students to develop strategies for working problems in a series of logical steps The Examples and Exercises give plenty of confidence building practice the end of chapter problems test the student's mastery The system of objectives tells the students exactly what they must learn in each chapter and where to find it **Introductory Chemistry** Kevin Revell, 2020-11-17 Introductory Chemistry creates light bulb moments for students and provides unrivaled support for instructors Highly visual interactive multimedia tools are an extension of Kevin Revell's distinct author voice and help students develop critical problem solving skills and master foundational chemistry concepts necessary for success in chemistry *Introduction to Chemistry* Bauer, Richard C. Bauer, Birk, Pamela Marks, 2015-03 Chemistry 2e Paul Flowers, Klaus Theopold, Richard Langley, Edward J. Neth, William R. Robinson, 2019-02-14 Chemistry 2e is designed to meet the scope and sequence requirements of the two semester general chemistry course The textbook provides an important opportunity for students to learn the core concepts of chemistry and understand how those concepts apply to their lives and the world around them The book also includes a number of innovative features including interactive exercises and real world applications designed to enhance student learning The second edition has been revised to incorporate clearer more current and more dynamic explanations while maintaining the same organization as the first edition Substantial improvements have been made in the figures illustrations and example exercises that support the text narrative Changes made in Chemistry 2e are described in the preface to help instructors transition to the second edition An Introduction to the Chemistry of the

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available in the ebook version *Chemistry* Bruce Averill, Patricia Eldredge, 2007 Emphasises on contemporary applications and an intuitive problem solving approach that helps students discover the exciting potential of chemical science This book incorporates fresh applications from the three major areas of modern research materials environmental chemistry and biological science Introduction to Atmospheric Chemistry Daniel J. Jacob, 1999 Atmospheric chemistry is one of the fastest growing fields in the earth sciences Until now however there has been no book designed to help students capture the essence of the subject in a brief course of study Daniel Jacob a leading researcher and teacher in the field addresses that problem by presenting the first textbook on atmospheric chemistry for a one semester course Based on the approach he developed in his class at Harvard Jacob introduces students in clear and concise chapters to the fundamentals as well as the latest ideas and findings in the field Jacob s aim is to show students how to use basic principles of physics and chemistry to describe a complex system such as the atmosphere He also seeks to give students an overview of the current state of research and the work that led to this point Jacob begins with atmospheric structure design of simple models atmospheric transport and the continuity equation and continues with geochemical cycles the greenhouse effect aerosols stratospheric ozone the oxidizing power of the atmosphere smog and acid rain Each chapter concludes with a problem set based on recent scientific literature This is a novel approach to problem set writing and one that successfully introduces students to the prevailing issues This is a major contribution to a growing area of study and will be welcomed enthusiastically by students and teachers alike Prentice Hall Chemistry Antony C. Wilbraham, 2006-10-15 Prentice Hall Chemistry meets the needs of students with a range of abilities diversities and learning styles by providing real world connections to chemical concepts and processes The first nine chapters introduce students to the conceptual nature of chemistry before they encounter the more rigorous mathematical models and concepts in later chapters The technology backbone of the program is the widely praised Interactive Textbook with ChemASAP which provides frequent opportunities to practice and reinforce key concepts with tutorials that bring chemistry to students through Animations Simulations Assessment and Problem solving tutorials

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of critical thinkers to ask questions to apply rules and models and to **Introductory Chemistry** Charles H. Corwin,2013-01-02 This edition features the exact same content as the traditional text in a convenient three hole punched loose leaf version Books a la Carte also offer a great value for your students this format costs 35% less than a new textbook With an expanded focus on critical thinking and problem solving the new Seventh Edition of Introductory Chemistry Concepts and Critical Thinking prepares students for success in Introductory Chemistry courses Unlike other introductory chemistry texts all materials the textbook student solutions manual laboratory manual instructor s manual and test item file are written by the author and tightly integrated to work together most effectively Math and problem solving are covered early in the text Corwin builds student confidence and ability through innovative pedagogy and technology formulated to meet the needs of today s learners By presenting chemistry in a clear and interesting way students to leave their first chemistry course with a positive impression a set of new skills and the desire to learn more Package consists of Books a la Carte for Introductory Chemistry Concepts and Critical Thinking 7 e **Chemistry, Life, the Universe and Everything** Melanie Cooper,Michael Klymkowsky,2014-06-27 As you can see this molecular formula is not very informative it tells us little or nothing about their structure and suggests that all proteins are similar which is confusing since they carry out so many different roles **Introduction to Chemistry** JAMES. BAUER BIRK (RICH. MARKS, PAMELA.),BAUER,Pamela S. Marks,2018-01-03

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1. This book is structured into several chapters, namely:
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 - Chapter 2: Essential Elements of Introduction To Chemistry Answer Key
 - Chapter 3: Introduction To Chemistry Answer Key in Everyday Life
 - Chapter 4: Introduction To Chemistry Answer Key in Specific Contexts
 - Chapter 5: Conclusion
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