A saturated solution of NaNO3 is prepared at 60.°C using 100. grams of water. As this solution is cooled to 10.°C, NaNO3 precipitates (settles) out of the solution. The resulting solution is saturated. Approximately how many grams of NaNO3 settled out of the original solution?



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hydration of the solid First the sparingly soluble anhydrous beryllium magnesium and calcium hydroxides whose freshly precipitated solids are poorly crystalline and show decreasing solubility with aging and whose solubility in water decreases with increasing temperature Second the soluble strontium and barium hydroxide octahydrates that form crystalline precipitates which do not show changes in solubility on aging and whose solubility in water increases with increasing temperature

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A Saturated Solution Of Nano3 Is Prepared At 60 Introduction

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